LA-5106-PR

PROGRESS REPORT

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Quarterly Progress Report on the Advanced Plutonium Fuels Program

July 1 to September 30, 1972



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This report presents the status of the LASL Advanced Plutonium Fuels program. The four most recent reports in this series, all unclassified, are:

LA-4841-MS	LA-4993-PR
LA-4913-PR	LA-5067-PR

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Advanced Plutonium Fuels Program

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ABSTRACT

This is the 24th quarterly report on the Advanced Plutonium Fuels Program at the Los Alamos Scientific Laboratory.

Most of the investigations discussed here are of the continuing type. Results and conclusions described may therefore be changed or augmented as the work continues. Published reference to results cited in this report should not be made without obtaining explicit permission to do so from the person in charge of the work.

PROJECT 401

EXAMINATION OF FAST REACTOR FUELS

Person in Charge: Principal Investigators: J. W. Schulte

R. D. Baker K. A. Johnson

G. R. Waterbury

INTRODUCTION I.

This project is directed toward the examination and comparison of the effects of neutron irradiation on LMFBR Program fuel materials. Unirradiated and irradiated materials will be examined as requested by the Fuels and Materials Branch of DRDT. Capabilities are established and are being expanded for providing conventional preirradiation and postirradiation examinations. Nondestructive tests will be conducted in a hot cell facility specifically modified for examining irradiated prototype fuel pins at a rate commensurate with schedules established by DRDT.

Characterization of unirradiated and irradiated fuels by analytical chemistry methods will continue, and additional methods will be modified and mechanized for hot cell application. Macro- and micro-examinations will be made on fuel and cladding using the shielded electron microprobe, emission spectrograph, radiochemistry, gamma scanner, mass spectrometers, and other analytical facilities. New capabilities will be developed in: gamma scanning, analyses to assess spatial distributions of fuel and fission products, mass spectrometric measurements of burnup and fission gas constituents, chemical analyses, and measurement of carbon in irradiated fuels.

Microstructural analyses of unirradiated and irradiated materials will continue using optical and electron microscopy, and autoradiographic and x-ray techniques. Special emphasis will be placed on numerical representation of microstructures and its relationship to fabrication

and irradiation parameters. New etching and mounting techniques will be developed for high burnup materials.

EQUIPMENT DEVELOPMENT Π.

Α. Inert Atmosphere Systems (P. A. Mason, E. O. Quintana)

The shaft seals on the recirculating pump of the Disassembly Cell purifier deteriorated during September requiring isolation of the unit from the cell. An inert atmosphere has been maintained by an Ar once-through purge. Typical concentrations of < 10 ppm O_2 and < 5 ppm H_2O are being maintained with a 60 cfh purge. Replacement seals have been procured, with installation expected to be completed by October 13.

The "DTA Cell" was "inerted" over the weekend of September 24 to permit the introduction and handling of sodium. Typical concentrations of < 25 ppm O_2 and < 10 ppm H₂O were achieved.

Reduction of the air leak rates to the metallography cells was achieved by the installation of Model "L" Manipulators. Concentrations of < 25 ppm O_2 and < 10 ppm $H_{0}O$ have been obtained on occasions. The large flow rates through the "slurper" solvent disposal system have, on several occasions, exceeded the makeup flow rate capability of the alpha-box pressure differential control system, resulting in excessive negative pressure in the alpha-box. The negative pressure dropped to the level where air was admitted through the oil trapped, over-pressurization protection device. A modification of the makeup Ar gas flow control system to permit higher flow rates was made, and

instructions were given to the cell operators regarding the proper use and control of the "slurper" system.

Preliminary evaluation testing has begun on the recently purchased Vacuum/Atmospheres MO-40 Gas Purifier System. The unit is being prepared as a mobile backup purifier to the present recirculating purifier units and for temporary installation at other inert cells as required. B. Manipulator Maintenance

Considerable difficulty is being encountered with the frequent "breakdown" of Model "L" manipulators. An average of one manipulator repair per week has been required during this report period. The load capacity of this type manipulator is approximately one-third that of the AMF Standard Duty Model 8 manipulators.

Some in-cell equipment has been modified to reduce the manipulator force requirements. Careful use of the manipulators by Operators has also been stressed.

An additional PaR Motion Mini-Manipulator was received in August along with a number of spare parts needed for proper maintenance.

C. Fuel Pin Handling System for Betatron Radiography (C. D. Montgomery, T. Romanik, J. R. Trujillo)

Shop fabrication of the handling mechanism is now complete, and it is in operational check-out phase prior to final installation. Some delay was experienced in delivery of the ball screw assembly and the pneumatic tire wheels.

All components of the new shielded radiography cask were fabricated and inspection has been completed. Final assembly will start the first week of October 1972. It is anticipated that this cask will be ready for local use by the middle of November.

A special indexing base in the concrete floor of the Betatron building to provide exact positioning of the cask for radiography of fuel pins will also be completed by mid-November.

Work is still in progress to obtain Special DOT approval for using this cask as an off-site shipping container.

- D. Cask Insert Elevator
 - (J. M. Ledbetter, C. D. Montgomery, D. C. Maestas, J. B. Weber)

The second Rover Cask insert elevator was completed and tested in operation with a 275-lb load. It has since been used satisfactorily in loading and unloading fuel from a Rover Cask.

E. Butyl Acetate Removal System

(G. S. Dow, M. E. Lazarus, P. A. Mason)

All of LASL's work has been completed on this system, and a job order has been submitted to the crafts for the remainder of work and installation of the system.

F. Storage Jib for 25-Ton Cask Sling

(C. D. Montgomery, T. Romanik, J. R. Trujillo) A special jib crane for storing the 25-ton, fourbranch sling used to lift the 21-ton Rover Casks at the DP West Facility has been designed. This device will receive the sling from the hook of the 25-ton hoist and swing it out of the way when not in use. It will also deliver the sling to the hoist hook as needed through manual manipulation.

This item will be completed and installed by mid-November.

G. In-Cell Equipment

(M. E. Lazarus, T. Romanik)

1. Optical Gauge for Diameter Measurement. The modification of the B and L DR25B optical gauge has been completed and preliminary tests indicate that the system works properly. Installation of the gauge at the DP-West Facility is now in progress.

2. Fuel Element Length Gauge. Thermocouple systems have been designed which allow temperature measurements of fuel elements with 300 series stainless steel claddings as well as other cladding materials.

The gauge is constructed of 304 stainless steel and can measure materials with the same coefficient of expansion without requiring temperature corrections, provided the material and the gauge are at the same temperature.

3. Mechanical Profilometer. The mechanical profilometer will be used primarily for breached fuel elements. Since it may have a low use rate, the replacement profilometer will have only a few modifications over the one previously installed in the Disassembly Cell. The new unit, which is now under construction, will allow easier maintenance and will eliminate bow error, if required, by physically aligning the fuel element. Construction is now 50% complete.

<u>4. Electro-Optical Profilometer</u>. A new Electro-Optical unit has been ordered which should completely eliminate the error produced when the fuel element moves off center. This unit puts out a digital signal directly instead of an analog signal. The advantage of a direct digital signal is that the noise level is much lower than in an analog system. Since the repair time on the old unit could be as much as 4 to 6 months, it will be regarded as a spare when the new unit has been installed.

5. <u>Macrophotography</u>. The macrophotography system is ready for use. However, it will be modified to include placing more legible numbers on the fuel element position device to indicate position along the length. This will aid in improved documentation on the photographs.

III. MICROSTRUCTURAL ANALYSIS

(J. H. Bender, D. D. Jeffries, K. A. Johnson, J. L. Lehmann, H. D. Lewis, K. L. Walters)

A. Etching

Ion gun etching is still being tested for maximation of etching conditions and for remote operation.

B. Image Analyses

Several modifications have been incorporated into the CDC 7600 computer program IMAGE to provide output formats which are suitable for direct transmittal of reduced QTM image analysis data to the experimenter. The new modules acquired for the Quantimet have made possible several additional modes of image analysis, one of these being the measurement of the sample distribution of sizes of planar areas. Examination of the standard Saltykov method for analysis of this type of data has led to the development of more general methods for practical data analysis. A detailed discussion is being prepared for journal publication.

IV. INSTALLATION OF AUTOMATIC PHOTOGRAPHIC EQUIPMENT

(P. W. Montoya, L. A. Waldschmidt)

The new automatic Kodak processor and printer have been installed. Testing and servicing by Kodak Engineers will be completed by November 10 at which time LASL will assume responsibility for operating the equipment.

A Technician was sent to the Kodak Training Center in Rochester, N.Y., for a training program in the use and maintenance of the specialized Kodak 5B-K Printer.

V. HOT CELL FACILITY AT DP-WEST

(M. E. Lazarus, C. D. Montgomery, J. R. Trujillo) The DP-West Hot Cell Facility is now considered operational. Any future reporting of modification to equipment and building services will be listed in pertinent sections of the Quarterly Reports. A view of the Operating Area after installation of equipment is shown in Figure 401-1.



Fig. 401-1. View of Operating Area at the DP-West Facility

VI. ANALYTICAL CHEMISTRY

1. Gamma Scanning.

(J. R. Phillips, G. H. Mottaz, J. N. Quintana, J. R. Netuschil)

a. Two-Dimensional Radial Isotopic Distributions.

Nondestructive gamma scanning was used to determine the two-dimensional, radial isotopic distributions of three fission products at three axial locations in an irradiated fuel pin containing $(U, Pu)O_2$ fuel. The distributions of 106 Rh, 137 Cs, and 95 Zr were determined at each axial position by scanning the fuel pin at 0° and 90°, and unfolding the data using the TWODIM computer code. ¹ The computer code accepts one or more diametral scans taken at different angles around the outside of the fuel pin and determines the distribution of the measured isotope over the cross section. The results of these measurements serve as an example of the type of information obtainable by this relatively new technique. The diametral isotopic distribution of 106 Rh shown at the top of Fig. 401-2 was used to generate the density plot shown in the bottom of the figure. Similarly, the typical diametral distributions of 137 Cs and 95 Zr with their corresponding density plots are shown in Figs. 401-3 and 401-4, respectively. The 95 Zr distribution also is presented as an isometric projection in Fig. 401-5.







Fig. 401-3. ¹³⁷Cs diametral isotopic distribution (Top) with two-dimensional isotopic distribution (Bottom).

From these examples.can be deduced the probable migrations of these three fission products which were chosen because of their different properties. The ¹⁰⁶Rh (Figure 401-2) concentrated radially between the central void and the columnar grain boundary. The ¹³⁷Cs migrated radially, concentrating near the outer surface. The ⁹⁵Zr remained with the fuel material, showing the relative position of the fuel. This nondestructive technique does not





require sectioning of the sample to provide the two-dimensional isotopic distributions of fission products in irradiated fuel pins.

b. Equipment at DP West. The DP West computer-controlled, gamma scanning system is in operation, having scanned six fuel pin sections. The data processing computer code, SURVEY, was modified, by incorporating additional plotting routines, to increase data manipulation capabilities.



ZR 95 724 KEV 25 995 IN

Fig. 401-5. Isometric projection of the ⁹⁵Zr isotopic distribution shown in Fig. 401-4.

2. Determination of U and Pu in Irradiated Fuels. (J. W. Dahlby and R. R. Geoffrion)

Controlled-potential coulometry provides the capability to measure U and Pu without chemical separation from each other in solutions of irradiated (U, Pu)O2 fuels. The U is measured by integrating the current while reducing U(VI) to U(IV), and subtracting a blank obtained by again reducing the same sample after a 6-min waiting period. The Pu is determined by integrating the current to oxidize Pu(III) electrolytically to Pu(IV) and then reducing the Pu(IV) to Pu(III). This oxidation-reduction cycle is repeated until successive integrated currents agree within 2 mV or 2 μ g of Pu. Neither separation of the U and Pu from each other nor from the fission products is required for samples having up to 6% burnup. Small samples containing approximately 1 mg of fuel can be analyzed for U and Pu with a precision of 0.5% relative standard deviation.

The analysis of fuels having burnups greater than 6% is adversely affected by the increased radioactivity and fission products. Because of this, a separation of the U and Pu from the highly radioactive fission products is

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being investigated. Precipitation separations are quick and easy to perform under remote conditions so they are being investigated. In the most promising precipitation separation method, the U and Pu are precipitated with NH_4OH , filtered, and the U is selectively dissolved in a basic solution of NH_2OH . Greater than 95% of the β - γ active fission products were removed from the U using this separation scheme. Recoveries of U were 99.6%. Most of the 0.4% loss was caused by a small amount of U (< 1 mg of the original 200 mg of U) not dissolved by the NH_2OH when Pu is present. The U is quantitatively dissolved if Pu is absent.

Several techniques are being tried to improve the dissolution and recovery of U without reducing the fission product removed. Only slight improvement in the U recovery from 99.6% to 99.7 to 99.8% has been attained.

VII. REQUESTS FROM DRDT

A. Examination of Unirradiated Fuel Pins (M. E. Lazarus)

Measurements were made on the Optical Profilometer of 20 BMI and 7 ORNL fuel pins. These pins are being followed by LASL personnel.

 <u>B.</u> Examination of Irradiated Materials

 (R. N. Abernathey, K. A. Johnson, M. E. Lazarus, R. A. Morris, J. R. Phillips, J. W. Schulte, G. R. Waterbury, W. F. Zelezny)
 <u>General Electric Company:</u> Examinations per

formed on irradiated fuel assemblies received on February 14, 1972 are tabulated in Table 401-I.

TABLE 401-I

POSTIRRADIATION EXAMINATIONS OF PINS FROM GE

Examination	Pin Identity				
Sectioning	F12P, F12Q, F4A, F4E				
Photography (Sectioned Faces)	F4A, F4E				
Length Measurement	F4A, F4E				
Density of Pin Cladding	F4A, F4E, F9C-13				

Fuel and clad specimens from GE-F9C-13, GE-F4A, and GE-F4E were dissolved, and burnup measurements were made.

The shielded electron microprobe was used for examining a cross-section sample of GE-F9C-13 (F). Three clad samples from each fuel pin (GE-F9C-13, GE-F4A, and GE-F4E) were cleaned and prepared for density measurement.

Microstructural examinations consisting of macrophotography, alpha and beta-gamma autoradiography, and optical microscopy (including mosaics) were carried out in an Ar atmosphere on specimens as listed in Table 401-II.

TABLE 401-II

MICROSTRUCTURAL ANALYSIS OF GE MATERIALS

<u>GE Pin No.</u>	No. Samples Fuel and Clad	Section Longit. or	n Type Transverse
GE-F9C-13	. 2	1L	1T
GE-F4A	2	1L	1T
GE-F4E	2	1L	1T
GE-F12P	3	2L	1T
GE-F12Q	3	2L	1T

Two samples were also prepared for electron microprobe including pre- and post-EMX photomicroscopy. Three samples were also analyzed using image analyses; these reports are in progress.

<u>Gulf United Nuclear Fuels Corporation</u>: Examinations of irradiated fuel pins received February 22, 1972 are tabulated in Table 401-III.

Hanford Engineering Development Laboratory: Examinations of irradiated fuel pins received February 22, 1972 are tabulated in Table 401-IV.

TABLE 401-III

POSTIRRADIATION EXAMINATION OF GU MATERIAL

Examination	Pin Identity			
Profilometry (Electro-Optical)	UNC-138, UNC-146			
Photography	UNC-194			
Eddy Current	UNC-194			

TABLE 401-IV

POSTIRRADIATION EXAMINATIONS OF PINS FROM HEDL

Examination	Pin Identity				
Profilometry (Electro-Optical)	P-17A-5, -19, -20, -29, -31 and -33				
Sectioning	P-17A-16, -17, -19, -20 and -27				
	- •- • ·				

Gamma scanning, including 7 gross and 6 complete spectral scans, was applied to the nondestructive

examination of HEDL-P17A-16. Also on this same fuel pin, radial isotopic distributions of ¹⁰⁶Rh, ¹³⁷Cs, and ⁹⁵Zr were determined at 3 axial positions using the TWO-DIM computer code to unfold the individual fission product distributions.

Cross-section samples from HEDL-P-17A-16 (H), HEDL-P-17A-2 (D), and HEDL-P-17A-26 (B) were examined using the shielded electron microprobe.

A cross-section sample of fuel and clad from HEDL-P-17A-27 was dissolved and analyzed for burnup.

Microstructural examinations consisting of macrophotography, alpha and beta-gamma autoradiography, and optical microscopy (including mosaics) were carried out in an Ar atmosphere on specimens as listed in Table 401-V.

TABLE 401-V

MICROSTRUCTURAL ANALYSIS OF HEDL MATERIALS

<u>HEDL Pin No.</u>	No. Samples Fuel and Clad	Section Longit. or	on Type <u>Transverse</u>
HEDL-P17A-18	5	2L	3Т
HEDL-P17A-26	3	2L	1T
HEDL-P17A-27	3	2L	1T

Three samples were also prepared for electron microprobe including pre- and post-EMX photomicroscopy.

Los Alamos Scientific Laboratory: The fuel was removed from 2 fuel-clad sections of pin 42-B. The clad sections were processed in the metallography cells. The resulting polished specimens, in 1.25-in. mounts, will be shipped to ANL-East for examination.

Oak Ridge National Laboratory: The fission gas from ORNL-43-N2-1, -2, -3, and -4 was analyzed mass spectrometrically.

Other Material From Experimenters: The schedule for examining materials at LASL indicates that approximately 134 pins will be received by May 1, 1973. This listing was determined after discussions with Sponsors from ANL, GE, HEDL, and LASL. Distribution of the pins is tabulated in Table 401-VI.

This partial list of pins may be modified by changes in EBR-II Reactor Schedule, priority adjustments by RDT, or changes in plans of the various Experimenters.

TABLE 401-VI

FUEL PINS TO BE EXAMINED AT LASL DURING FY 1973-1974

Sponsor	N	o. of Pins
ANL		1
GE		11
HEDL		84
LASL		38
	Total	134

VIII. PUBLICATIONS, TALKS

- C. D. Montgomery, M. G. Chavez, M. E. Lazarus, "Development of Ancillary Equipment for Usc in Hot Cell Facilities," Proceedings of the 20th Conference on Remote Systems in Technology, American Nuclear Society, 331 (1972).
- D. D. Jeffries, J. H. Bender, and K. A. Johnson, "Improvements in Remote Metallography in Incrt Atmospheres," Proceedings of the 20th Conference on Remote Systems Technology, American Nuclear Society, 143 (1972).
- J. R. Phillips, G. R. Waterbury, G. H. Mottaz, and J. N. Quintana, "New Systems for Gamma Scanning Fuel Elements," Proc. 20th Remote Systems Technology, American Nuclear Society, 115 (1972). Also presented as a talk at the meeting, Idaho Falls, ID, September 19-21, 1972.
- C. S. MacDougall, M. E. Smith, and G. R. Waterbury, "Remotized Apparatus for Dctermining Oxygen in Irradiated Reactor Fuels and Cladding Materials." Presented at the 20th Conference on Remote Systems Technology, American Nuclear Society, Idaho Falls, ID, September 19-21, 1972

IX. REFERENCES

 B. K. Barnes and J. R. Phillips, "TWODIM, A Computer Code for Unfolding Diametral Gamma-Ray Scans on Reactor Fuel Elements," Los Alamos Scientific Laboratory Report LA-4676 (May 1971).

CORRECTION -- PROJECT 401

In LA-5067-PR, p. 8, left-hand column, 2nd sentence, should read as follows: "To determine if the cesium activity could be related to the oxygen contents of the specific pellets, analyses were made for total oxygen in pellets having high cesium activity and in pellets with low cesium activity. The preliminary results showed that the cesium activity was directly related to the oxygen contents."

PROJECT 463

HIGH PERFORMANCE LMF BR FUEL MATERIALS

Person in Charge: R. D. Baker Principal Investigator: J. L. Green

I. INTRODUCTION

The primary objective of this program is the overall evaluation of the most promising of the candidate fuel systems for advanced LMFBR application. Emphasis currently is placed on the study of the relative merits of stainless steel clad nitride and carbide fuels under conditions that appropriately exploit the potential of these materials to operate to high burnup at high power densities. The major portion of the program is the evaluation of the irradiation performance of these fuel element systems. A continuing series of irradiation experiments is being carried out under steady state conditions in fast reactor environments to assess the effects of damage and burnup on stainless steel clad, carbide and nitride fuel elements. These experiments are designed to investigate fuel swelling, interactions between the fuel and clad and thermal bonding medium, fission gas release, and the migration of fuel material and fission products as a function of burnup and irradiation conditions. In addition, experiments are being designed to allow the study of the effects of rapid, overpower, reactor transients on car bide and nitride fuel assemblies. Contiguous efforts are necessary in the development of fuel material preparation and fabrication procedures as well as the techniques required for the characterization of fuel materials both before and after irradiation.

A second objective in the program is the determination of thermophysical, mechanical and chemical properties and characteristics of plutonium-containing ceramics that are required for their evaluation and use as fuel materials. A broad range of capabilities in this area has been developed, including the study of (1) phase relationships using differential thermal analysis, (2) thermal transport, (3) thermal stability and compatibility, (4) hot hardness and its temperature dependence, (5) structure and phase relationships using high temperature x-ray and neutron diffraction, (6) thermal expansion, and (7) compressive creep rates as a function of temperature and stress. Several of these techniques are available for use with irradiated fuels.

II. IRRADIATION TESTING

The objective of the irradiation testing program is the overall evaluation of the most promising of the candidate fuel systems for advanced LMFBR application. The irradiation experiments are carried out under conditions that take advantage of the potential of these materials to operate to high burnup at high power densities.

A. Synthesis and Fabrication

(K. W. R. Johnson, H. Moore, C. Baker, R. Walker, C. W. Bjorklund, and J. G. Reavis)

<u>1.</u> Carbide Production. Quality Assurance requirements make it necessary to have complete fuel specifications before pellets can be produced. Dimensional parameters for upcoming EBR-II and Treat experiments have not been finalized; hence, no fuel pellets have been produced. At the time the QA program was put into effect, four batches of pellets were in process. Of these four batches, two were in the form of annealed ingots which are being stored for future use. One batch of H₂ treated powder was stored in a vacuum sintering furnace. This material deteriorated due to repeated removal of the powder from the system when it was necessary to use the furnace for other purposes. A second batch of H_2 treated powder was stored in a vacuum-tight container. This material was sampled for O_2 after a period of 3 months and it was found to contain ≤ 75 ppm.

2. Process Development

a. Carbide Production by Carbothermic Reduction of UO₂-PuO₂

Development of the carbothermic reduction process for production of pure (low oxygen), single phase $U_{0.8}Pu_{0.2}C$ is continuing. The steps used in this process are:

1. Ball mill and blend UO_2 , PuO_2 and excess graphite,

2. Form low density powder compacts,

3. Heat in vacuum at 1400-1700⁰ until CO evolution ceases,

4. Crush and comminute,

5. Heat in flowing H_2 at 800[°]C to remove higher carbide,

6. Press to form fuel pellets,

7. Sinter under Ar.

It was reported previously¹ that $U_{0.8}Pu_{0.2}C$ containing acceptably low oxygen concentrations were prepared on the 50 g scale. Recent experiments have been run to demonstrate the process on the 250 g scale. These experiments have also had as their aims the optimization of carbon concentration in the oxide-graphite mixture, and selection of optimum reaction conditions.

Historically, the two most serious problems encountered in the preparation of single-phase monocarbides using established carbothermic reduction techniques have been high oxygen concentrations in the product and the control of the carbon content of the product. This new procedure has been devised to eliminate these two problems. The unique features of the LASL process are:

1. The initial oxide-carbon mixture is intentionally made carbon rich to ensure the presence of sesquicarbide in the reduction product. This is done to maintain a large carbon activity during the final stages of the reduction. 2. The final high temperature stage of the carbothermic reduction step is carried out in high vacuum. This is critical to the preparation of material having low residual oxygen contents.

3. The use of the LASL hydrogen reduction procedure to remove all higher carbides from the product.

Carbon concentrations used and results of analyses of the products of some of these experiments are listed in Table 463-I. Comparison of the data from experiments CR-8 and CR-9 seems to show that a trade-off must be made between oxygen concentration and M_2C_3 concentration in the reduction product, and that these concentrations can be effectively regulated by varying the graphite content of the original mixture. The results of CR-10 show that low oxygen concentrations can be maintained in the product when the process is operated on the 250 g scale. The difference in procedure believed to account for the differences in oxygen concentrations of the products of CR-10 and CR-11 is that Batch CR-10 was held at 1700[°] for 130 min during reduction, while Batch C-11 was held at 1700[°] for 110 min.

The product of CR-10 was taken through the remaining steps of the process. The fuel pellets produced contained traces (< 1 vol%) of platelets which indicated that a somewhat longer H₂ reduction treatment is needed.

The oxygen concentration in the final pellets was 330 ppm. The product of reduction CR-11 has not yet been taken through the final steps of the process.

TABLE 463-I							
SUMMARY OF RECENT C	ARBOTHERMIC	REDUCTIONS					
FOR PRODUCT	ION OF U. Pu	_,C					

Experiment No.	Batch Size, g	Original C Conc, wt%	O in Product, ppm	M ₂ C ₃ in Product, vol [%]
CR-8	50	12,12	30	~ 15
CR-9	50	11.92	475	< 1
CR-10	250	11.99	10	~ 5
CR-11	250	11.99	290	~ 5

^aProduct from the carbothermic reduction step.

<u>b.</u> Pellet Production and Characterization. The density specification for carbide fuel for the Series K4 tests has been set at $95 \pm 2\%$ TD. Batches 37 and 38 were 92.9 and 92.5% TD. Both batches were resintered at 1850° C for 8 h and then at 1400° C for 2 h. The densities increased to 95.0 and 94.0, respectively, with no alteration in the chemical composition. Other batches which have densities below the specification are being resintered.

During the examination of a storage container of archival fuel it was observed that three adjacent pellets showed evidence of deterioration. This material was subjected to chemical and microstructural analysis. The chemical analysis indicated that oxygen was the only contaminant which was present in a larger amount (0.16 wt%) than that reported for the as-fabricated pellets. Microstructural analysis showed a structure replete with microcracks, typical of room temperature oxidation. Further investigations are being carried out in an attempt to identify the source of the contaminant.

3. Equipment Development. A furnace similar to the new hydrogen treatment furnace¹ was installed in the nitride preparation glovebox. Each of these furnaces, when heated in an inert atmosphere caused an excessive temperature increase inside of the glovebox. The furnace heat loss was decreased substantially by adding insulation to the furnace. It was observed that after insulating the furnace in the nitride glovebox the atmosphere temperature remained at an acceptable level. The atmosphere recirculation unit for this glovebox incorporates a water cooled heat exchanger which provides sufficient cooling to maintain the glovebox temperature at an acceptable level during furnace operation. A similar heat exchanger is being installed in the glovebox housing the H_o treatment furnace.

A high temperature, tungsten mesh, resistance heated, sintering furnace is being installed in a recirculating, inert atmosphere glovebox. The completed installation will provide a temperature capability of 2750° C in atmospheres of N₂, Ar, He, H₂ or vacuum. Temperature cycles may either be controlled manually or programmed. The furnace has been leak tested and operated at 1400° C and 8×10^{-7} torr. In anticipation of the tight fuel pellet diameter tolerances that will be required for fuel elements using close fitting shroud tubes, a centerless grinder has been fabricated. An existing, recirculating, inert atmosphere glovebox is being installed to house the grinding operation. The box was previously used for nonradioactive materials, so glove port and exhaust system modifications are neccssary to convert it to use with plutonium.

An electronic balance was acquired for use in the fuel preparation facility. Preliminary evaluation indicated that the balance did not meet specifications or the program needs, hence it was returned to the manufacturer for modification and refinement.

4. Nitride Fuel Pellet Evaluation. A total of $258 U_{0.8}Pu_{0.2}N$ pellets comprising four batches of fuel were received from Battelle Memorial Institute. Radiographic examination indicated that 170 pellets were either cracked or chipped. Further chacterization will be carried out after the BMI QA documentation is received for this material.

B. EBR-II Irradiation Testing

(J. O. Barner, T. W. Latimer, L. L. Marriott, H. E. Strohm)

The purpose of the EBR-II testing program is the evaluation of the steady-state irradiation behavior of high performance fuel element systems for application in advanced LMFBR reactors. Several series of carbide- and nitride-fueled experiments have been initiated in the past several years. The main objectives of the irradiations are: (1) the development of fuel element designs for use with both fuel types; (2) the determination of the irradiation behavior of the fuel materials; (3) a comparison of sodium and helium bonding; (4) a comparison of different cladding alloys; and (5) the evaluation of the overall irradiation performance of the fuel element systems. The majority of the experiments under test or that have been completed have been encapsulated. Most of the experiments that are currently available for irradiation or that are being designed are singly clad.

Fourteen series of experiments have been originated. The description and status of these series are summarized in Tables 463-II to 463-XI.

Experiment	Series	Fuel Type ^b	Fuel Density, <u>% T. D.</u>	Diametral Gap, in.	Max. Fuel Temp. at <u>Startup, C</u>	Max. Linear Power, Kw/ft	Current Burnup at.%	Status
K-36B	1	(U _{0.8} Pu _{0.2})C	90	0.015	1165	30	6.8	$NDT - out^{f}$
K-37B	1	(U _{0.8} Pu _{0.2})C	90	0.015	1165	30	3.2	NDT - out ^C
K-38B	1	(U _{0.8} Pu _{0.2})C	90	0.015	1165	30	6.4	X152 - in ^d
K-39B	1	(U _{0.8} Pu _{0.2})C	90	0,015	1165	30	6.4	X152 - in
K-40B	1	(U _{0.8} Pu _{0.2})C	95	0.020	1150	30		To be built
K-41B	1	(U _{0.8} Pu _{0.2})C	95	0.020	1150	30		To be built
K-42B	1	(U _{0.8} Pu _{0.2})C	90	0.015	1165	30	5.0	Completed ^e
К-43	3	(U _{0.8} Pu _{0.2})C	95	0.020	1150	30	6.1	X152 - in
K-44	3	(U _{0.8} Pu _{0.2})C	95	0.020	115 0	30	6.1	X152 - in
K-45	3	(U _{0.8} Pu _{0.2})C	95	0.020	1150	30	3.0	NDT – out^{f}
K-46	3	(U _{0.8} Pu _{0.2})C	95	0.020	1150	30	2.9	NDT - out^{f}
K-47	3	(U _{0.8} Pu _{0.2})C	95	0.020	1150	30		To be built
K-48	3	(U _{0.8} Pu _{0.2})C	95	0.020	1150	30		To be built
K-49	2	(U _{0.8} Pu _{0.2})C	95	0.020	1400	45 - 50	4.0	NDT – out^{f}
K-50	2	(U _{0.8} Pu _{0.2})C	95	0.020	1400	45 - 50	4.0	NDT – out^{f}
K-51	2	(U _{0.8} Pu _{0.2})C	95	0.020	1400	45 - 50	3.9	NDT - out^{f}

TABLE 463-II SERIES K1, K2, AND K3 CARBIDE EXPERIMENTS^a

All elements are clad in 0.300 in. o.d. x 0.280 in. i.d. Type 316SS. All are sodium bonded elements.

The series 1 and 3 experiments are fully enriched in 235 U. The series 2 experiments contain 97% 233 U. All fuel cis single-phase.

Capsule K-37B was damaged during reconstitution of X152 to such an extent that it cannot be irradiated further. d^{Failure} indicated by γ -scan.

Capsule K-38B was damaged during reconstitution of X152 at EBR-II. Further irradiation is planned.

e Reported in LA-4669-MS.

Failure indicated by γ -scan at EBR-II.

Table 463-II describes the K1, K2, and K3 series of tests. In these experiments single-phase, high-purity, uranium-plutonium monocarbide pellets are sodium bonded to Type 316 stainless steel cladding. In general, the operating linear power ratings of the capsules are high (approximately 30 Kw/ft). Three tests at very high power (> 40 Kw/ft) were included to determine the effect of high thermal stresses and high fuel temperatures on fuel element behavior. Indications of element failure were found at EBR-II in several experiments from this group, five in

subassembly X119B and one from X142, using γ -scanning for ¹³³Xe. Preliminary examinations of these experiments in the LASL hot cells indicate that the capsules are bowed as much as 3/8 in. Further nondestructive examination of the failed experiments is currently under way.

Table 463-III describes the Series U1300 experiments. These experiments contain two-phase, uraniumplutonium carbide fuel pellets which are helium bonded to either Type 316 stainless steel or Incoloy 800 cladding. Two methods for the accommodation of fuel swelling were

TABLE 463-III

SERIES U1300 CARBIDE EXPERIMENTS

						Fuel-		Max.		
	•	Fuel	a 1. 1	Clad	D	to-	Max. Fuel	Linear	Current	
Experiment	8	Density	Clad	Thickness,	Diametral	Ciad	lemp. at	Power	Burnup	Status.
<u>No.</u>	Fuel Type	<u>% T.D.</u>	Material	<u>in.</u>	Gap, in.	Bond	Startup, C	Kw/IL	at. yo	Status
U93	MC + 5 v/o M ₂ C ₃	84	31655	0.030	0,004	He	1750	18.0	11.1	NDT - out
U94	$MC + 5 v/o M_2C_3$	84	31655	0.015	0.007	Не	1680	21.9	10.7	NDT - out
U97	$MC + 5 v/o M_2C_3$	84	INC-800	0.030	0.004	He	1750	18.0	11.0	NDT - out
U9 8	MC + 5 v/o M ₂ C ₃	84	INC-800	0.015	0.007	Не	1680	21.9	10.6	$NDT - out^{c}$
U105	$MC + 5 v/0 M_2C_3$	84	INC-800	0.030	0.008	Не	1900	15.1	11.5	NDT – out
U106	MC + 5 v/o M ₂ C ₃	84	INC-800	0.015	0.009	He	1825	19.8	10.9	NDT - out ^C
U110	MC + 10 v/o M ₂ C ₃	99 ^b	INC-800	0.015	0.014	He	1960	21.9	10.1	NDT - out
U11 3	$MC + 10 \text{ v/o } M_2C_3$	99 ^b	INC-800	0.030	0.010	Не	1880	16.9	11.4	NDT - out
U114	MC + 10 v/o M ₂ C ₃	99 ^b	INC-800	0.015	0.007	He	1575	22.1	10.4	NDT - out ^c

 $^{a}M = (U_{0.85}^{Pu} _{0.15})$

^bCored pellet with nominal 0.080-in. diameter axial hole.

^oFailure indicated by γ -scan at EBR-II.

investigated in this series, i.e., the introduction of internal porosity by the use of either low-density solid fuel pellets or high-density cored pellets. These experiments reached their goal burnup of 10 at.% in subassembly X-142 after operation at moderate linear power ratings (approximately 20 Kw/ft). Indications of element failure for three experiments were found at EBR-II using γ -scanning for ¹³³Xe. These capsules are in transit to the LASL hot cells.

The Series U1950 experiments are described in Table 463-IV. In these experiments, either two-phase or single-phase carbide fuel is helium bonded to Type 304 or 316 stainless steel or to Incoloy 800 cladding. Fuel densities range from 75 to 99% of theoretical. These experiments are currently at about three-fourths of their goal burnup after operation at low linear power (10 to 15 Kw/ft). An interim examination is scheduled after EBR-II Run 58.

The Series U1930 and U1960 experiments are described in Table 463-V. Experimental parameters include fuel type, fuel density, bond type, and cladding type. The operating linear power ratings for the experiment are relatively high (30-35 Kw/ft). Nondestructive examination of the experiments listed in part A of Table 463-V was completed several months ago. The results of these cxaminations showed that fuel elements U194 and U200 had failed. Destructive examination of U194 is currently under way. Significant results that have been determined for U194 are:

 Only a very small amount, < 0.1 cm³ at STP, of fission gas was released from the element to the capsule plenum.

2. Approximately 2 cm³ of sodium were transferred from the element to the capsule.

3. No significant amount of fission gas was released from the fuel element during the de-encapsulation procedure, which included heating the experiment to melt the sodium in the capsule-element annulus.

4. A suspect area delineated by a stain on the cladding was observed about 8-5/8 in. from the bottom weld.

5. Several regions of cladding ovality were observed in the fuel region of the element. In only one region, from 8 to 9-1/2 in. from the bottom weld, were

TABLE 463-IV

SERIES U1950 CARBIDE EXPERIMENTS

Experiment No.	Fuel Type ²	Fuel Density <u>% T.D.</u>	Clad <u>Material</u>	Clad Thickness, in	Diametral Gap, in.	Fuel- To- Clad <u>Bond</u>	Max. Fuel Temp. at <u>Startup, C</u>	Max. Linear Power <u>Kw/ft</u>	Current Burnup at.%	Status
U129	$MC + 5 v/o M_2C_3$	84	316SS	0.022	0.016	He	1755	12.8	8.3	X055A - in
U130	$MC + 5 v/o M_2C_3$	7 5	316SS	0.022	0.010	Нe	1500	13.1	8.3	X055A - in
U131	$MC + 5 v/o M_2C_3$	84	316SS	0.022	0.010	Нe	1495	13.1	8.2	X055A - in
U132	$MC + 5 v/o M_2C_3$	84	316SS	0.022	0.010	He	1495	12.8	8.1	X035A - in
U133	$MC + 5 v/o M_2C_3$	84	316SS	0.022	0.010	He	1495	12.8	7.9	X055A - in
U134	$MC + 5 v/o M_2C_3$	84	316SS	0.022	0.010	He	1495	12.8	7.9	X055A - in
U135	$MC + 5 v/o M_2C_3$	84	INC-800	0.022	0.010	He	1475	12.8	8.2	X055A - in
U136	$MC + 5 v/o M_2C_3$	84	INC-800	0.022	0.010	He	1475	13.3	7.8	X055A - in
U137	$MC + 10 v/o M_{2}C_{2}$	99	316SS	0.022	0,010	He	1440	13.4	6.9	X055A - in
U138A ^b	$MC + 10 v/0 M_{p}C_{3}$	99	316SS	0.022	0.010	He	1440	14.8	3.3	X055A - in
U139	MC + 10 v/o M ₂ C ₃	99	INC-800	0.022	0.010	He	1440	14.8	7.0	X055A - in
U140	мс	93	IN C- 800	0.022	0.010	He	1460	13.9	7.5	X055A - in
U141	MC	93	316SS	0.022	0.010	He	1460	14.3	7.4	X055A - in
U142	MC	93	316SS	0.022	0.010	Нe	1460	14.5	7.5	X055A - in
U143	$MC + 10 v/o M_2C_3$	99 ^C	INC-800	0.022	0.010	He	1395	12.8	7.7	X055A - in
U14 4	$MC + 10 v/0 M_2C_3$	99 [°]	316SS	0.022	0.010	He	1395	13.1	7.8	X035A - in
U145	MC	93	304SS	0.015	0.030	Na	820	13.4	7.3	X055A - in
U146A ^b	MC + 10 v/o M ₂ C ₃	99	304SS	0.015	0.030	Na	810	13.7	3.3	X055A - in
U147	MC+ 10 v/o M ₂ C ₃	99	IN C-800	0,015	0.030	Na	810	14.2	7.4	X055A - in

 $^{8}M = (U_{0.85}Pu_{0.15})$

^bCapsules 138 and 146 were removed at 45,000 MWD/MT for TREAT testing. Duplicates replaced the originals.

^cPellets cored with nominal 0.080-in. diameter axial hole.

there peaks in the profilometry curves at 90[°] with respect to each other.

6. An eddy current test of the cladding did not indicate a failure point. An apparent increase in bond electrical conductivity occurred about 6 in. above the bottom weld.

7. A test, comprised of heating the element to $\sim 150^{\circ}$ C and pressurizing to 20 psig, resulted in neither loss of pressure nor evidence of sodium expulsion.

The results indicate: (1) a failure of the fuel element below the effective element sodium bond level that was present <u>after</u> the failure, and (2) a very small failure cross-sectional area and/or a failure area that is presently plugged. Metallographic examination of the lower weld, which could not be adequately examined either visually or with the eddy current test, and of the suspect region at 8-5/8 in. from the bottom weld are in progress in an effort to determine the point of failure and the nature of the failure. Because the amount of sodium loss from the element to the capsule indicates a partial unbonded condition during operation after the failure, metallographic examination of sections from the lower and upper regions of the fuel column will be made in order to determine if there was a significant difference in fuel operating temperature in these regions.

The experiments listed in part B of Table 463-V are currently being irradiated or are awaiting reinsertion into the reactor. The experiments listed in part C of Table 463-V were used as replacement pins in order to allow the irradiation to be continued to the desired burnup in the lead experiments. Only a cursory post-irradiaation examination is planned for these elements. The experiments listed in part D of Table 463-V are awaiting insertion into the reactor. Capsule U261 will be returned

TABLE 463-V

SERIES U1930 AND U1960 CARBIDE EXPERIMENTS

Experiment	Fuci Type ^a	Puel Density <u>% T.D.</u>	Clad <u>Matertal</u>	Clad Thickness, in,	Diametral Gap, in.	To- Clad <u>Bond</u>	Max. Fuel Teinp. at <u>Startup, ^oC</u>	Linear Power <u>Kw/ft</u>	Current Burnup 	Satus
U187	MC + 5 v/o M ₂ C ₃	84	316SS	0.020	0.007	Яo	1935	30.0	4.5	Destructive Exam
U189	$MC + 5 v/0 M_2C_3$	84	1NC-800	0.020	0.007	lie	1935	30.0	4.5	Destructive Exam ^C
U191	мс	93	3 04 SS	0.015	0.030	Na	1148	31.7	4.7	Destructive Exam ⁰
U192	мс	93	304SS	0.015	0.030	Na	1148	31.7	4.7	Destructive Exam
U194	MC + 10 v/o M ₂ C ₃	97	304SS	0.015	0.030	Na	1132	33.1	5.0	Destructive Exam ^C
U195	MC + 10 v/o M2C3	97	304SS	0.015	0.030	Na	1132	33.1	5.0	Destructive Exam
U197	MC + 10 v/o M2C3	97	INC-800	0.015	0.030	Na	1132	33.4	5,0	Destructive Exam ^C
U198	$MC + 10 v/0 M_2C_3$	97	INC-800	0.015	0.030	Na	1132	33.4	5.0	Destructive Exam ⁰
U200	MC + 5 v/o M2C3	84	304SS	0.015	0.008	He	2042	30.8	4.6	Dostructive Exam ^C
U206	MC+ 5 v/o M2C3	90	316SS	0.020	0.008	He	2084	31.5	4.7	Destructive Exam ^e
U208	MC + 10 v/o M2C3	97 ^b	31655	0.020	0.009	He	1912	31.9	4.8	Destructive Exam
U188	MC + 5 v/o M2C3	84	316SS	0.020	0.007	He	1935	30.0	8.7	X152 - in
U190	MC+5 v/o M2C3	84	INC-800	0.020	0.007	He	1935	30.0	8.8	X152 - in
U193	мс	93	304SS	0.015	0.030	Na	1148	31.7	8.7	X132 - in
U196	MC + 10 v/o M2C3	97	304 SS	0.015	0.030	Na	1132	32.6	8.6	X152 - in
U199	MC + 10 v/o M2C3	97	INC-800	0.015	0.030	Na	1132	33.5	5.0	Interim
U2 01	$MC + 5 v/o M_2C_3$	84	304SS	0.015	0.008	He	2042	30.0	4.5	Interim
U207	$MC + 5 v/o M_2 C_3$	90	316SS	0.020	0.00ა	He	2088	31.7	4.7	Interim
U209	MC+ 10 v/o M2C3	97 ^b	31655	0.020	0.009	Не	19 09	30.9	4.6	interim
U185	MC + 10 v/o M2C3	96	316SS	0.020	0.011	lie	2 195	30.0	3.0	out
U1 86	MC + 10 v/o M2C3	96	316SS	0.020	0.011	ife	2195	30.0	3.0	out
U2 02	MC + 5 v/o M2C3	84	316SS	0.010	zero	Нс	1270	31.7	2.8	out
U2 03	MC+5 v/o M ₂ C ₃	84	316SS	0.020	zero	Нс	1260	31.4	2.8	out
U2 04	MC + 10 v/o M ₂ C ₃	97 ⁰	316SS	0.010	zero	He	1131	32.2	2.9	ont
U205	$MC + 10 v/0 M_2C_3$	97 ⁰	316SS	0.020	zero	ilc	1124	31.9	2.9	out
U260	$MC + 10 v/0 M_2C_3$	98	316Ss ^d	0.015	0.015	ile	2590	34.1	1	
U261	MC + 10 v/o M2C3	98	31655 ^e	0.015	0.015	He	2590	34.1	}	At L'BR-II
U262	$MC + 10 v/0 M_2C_3$	98	INC-800	0.015	0.015	Hc	2590	34.1	(

^aM = (U_{0.85}^{Pu}0.15)

^bCored pellets with nominal 0, 080-in, diameter axial hole.

^oNeutron radiography, x-radiography, and de-encapsulation complete. Elements 194 and 200 had failed.

^d20% cold-worked.

^CEddy current test at EBR-II indicates capsule band discontinuity.

to LASL for rework of an apparent sodium bond defect in the capsule-element annulus.

Table 463-VI describes the Series B-1 and B-2 experiments. These capsules are fueled with single-phase, uranium-plutonium mononitride. All the elements in this series are sodium-bonded and clad with either Type 304 or 316 welded stainless steel tubing. Operating linear power ratings for the experiments are relatively high (25-35 Kw/ft). Capsules B-1-4 and B-2-5 have been examined using γ -scanning techniques for the detection of ¹³⁷Cs. and both elements are apparently intact. Destructive examination of capsule B-2-8 gave no indication of element failure after 3.6 at.% burnup. The results of the latter examination are reported in BMI-1925.² Further irradiation of B-1-4 and B-2-5 is planned. The remaining experiments from this series are in subassembly X152 which is scheduled for an interim examination at the end of EBR-II Run 58. B-1-1, B-2-1, and B-2-6 will have been irradiated to their goal burnup at the end of Run 58.

TABLE 463-VI

Experiment No.	Fuel Density, <u>% T.D.</u>	Smear Density, <u>%</u> T.D.	Cladding Wall Thickness, in.	Max. Linear Power, Kw/ft	Current Burnup at.%	Status
B-1-1	94.1	80.2	0.020	27.9	6.5	X152 - in
B-1-2	93.9	80.5	0.020	27.1	6.3	X152 - in
B-1-4	93.8	84.9	0.020	28.6	3.0	Nondestructive Exam
B-2-1	94.7	81.5	0.020	32.7	6.3	X152 - in
B-2-2	94.4	81.5	0.020	32.5	6.2	X152 – in
B-2-3	94.0	81.0	0.020	32.4	6.3	X152 - in
B-2-5	94.2	75.5	0.015	32.4	3.0	Nondestructive Exam
B-2-6	94.0	81.8	0.010	36.6	6.2	X152 - in
B-2-7	93.9	81.7	0.010	36.7	6.1	X152 - in
B-2-8	93.8	81.6	0.010	36.5	3.6	Complete

SERIES B-1 AND B-2 EBR-II NITRIDE EXPERIMENTS^a

^aAll capsules are sodium bonded. Capsule cladding is 0.375 in. o.d. x 0.335 in. i.d. Type 304 stainless steel. Element cladding for the B-1 experiments is welded 0.290 in. o.d. Type 304 stainless steel. Element cladding for the Series B-2 experiments is welded 0.315 in. o.d. Type 316 stainless steel. The fuel is $(U_{0,8}Pu_{0,2})N$.

The B-3 series of experiments are described in Table 463-VII. This nitride-fueled series is similar to the B-1 and B-2 series except that three helium bonded experiments are included and the average operating linear power ratings are slightly higher. Gamma-scans made at EBR-II for 133 Xe indicated that the four sodiumbonded elements have failed, while the three heliumbonded elements have not failed. Preliminary visual examination at LASL indicates that the four failed elements are bowed as much as 3/8 in. Nondestructive examination of the failed elements is currently under way.

The Series U5100 singly-clad experiments are described in Table 463-VIII. In this group, either singlephase or two-phase carbide fuel is sodium bonded to Type 304 or 316 stainless steel or to Incoloy 800. In seven of the elements, a shroud is incorporated primarily to test the retention of fuel fragments by close fitting tubes. A secondary objective of the shroud is to study the effectiveness of the shroud alloy as a carbon getter. Experiment U255 has been rejected due to a tungsten inclusion in the bottom weld. That experiment will be replaced with a stainless steel dummy element. Insertion of the experiments into the reactor is pending completion of the quality assurance data package by the Gulf United Nuclear Fuels Corporation, the original experimenter.

The C-5 series of singly-clad experiments is described in Table 463-DX. Single-phase nitride fuel is sodium bonded to 20% cold-worked Type 316 stainless steel cladding in all of the fuel elements in this group. Profilometry measurements of the C-5 series elements have been made using the same equipment that will be used for the postirradiation examination. The LASL nondestructive fissile assay results indicate that the ²³⁹Pu content of the pins is 3% lower than the values based on the nominal composition. This discrepancy may be resolved when the quality assurance documents are received from BMI, the original experimenter. Shipment of selected elements to EBR-II is pending receipt of these documents and a LASL review of the experiments from a quality assurance standpoint.

The O-N1 series of singly-clad experiments is described in Table 463-X. These experiments are similar to the C-5 series. The elements are fueled with $(U_{0.8}Pu_{0.2})$ N which is sodium bonded to 20% cold-worked Type 316 stainless steel cladding. These elements were received with the cladding in the fueled region in an

TABLE 463-VII

Experiment <u>No.</u>	Fuel Density, % T.D.	Bond <u>Type</u>	Smear Density, <u>%</u> T.D.	Cladding Wall Thickness, in.	Max. Linear Power, Kw/ft	Current Burnup at.%	Status
B-3-2	88.4	Na	82.6	0.015	37.7	3.1	NDT – out^{c}
B-3-3	91.3	Na	85.2	0.015	38.9	3.1	$NDT - out^{c}$
B-3-4	93.9	Na	85.4	0.015	38.9	3.1	NDT - out ^C
B-3-5	90.4	Na	84.8	0.015	41.5	3.1	NDT – out^c
B-3-6	94.8	Heb	85.5	0.020	34.2	3.0	NDT - out
B-3-7	88.5	He	87.9	0.020	34.2	3.0	NDT – out
B-3-8	89.8	Heb	85.5	0.020	32.4	2.9	NDT – out

SERIES B-3 EBR-II NITRIDE EXPERIMENTS

^aCapsule claddings are type 304 stainless steel, 0.375 in. o.d. x 0.335 in. i.d. Element claddings are 0.315 in. o.d. ^bPellets are annular with a 0.070-in. diameter hole.

^cFailure indicated by γ -scan at EBR-II.

oxidized condition. All elements were electropolished to remove the cladding oxidation. The average and maximum diameter decreases after electropolishing were 0.0003 and 0.0005 in., respectively. Three elements have been rejected because of large fuel chips in the sodium annulus. The diameters of the elements have been measured on the same profilometer that will be used after irradiation. Nondestructive fissile assay results agree well with those calculated from the nominal composition. The status of shipment of these experiments to EBR-II is the same as for the C-5 series. Four of the fuel pins from Series O-N1 will be irradiated with fifteen pins from Series C-5.

The tentative description of the K4 series of experiments is shown in Table 463-XI. One of the primary purposes of the K4 Series is to provide data for a critical comparison of the overall irradiation behavior of carbide and nitride fuel elements which have been irradiated under conditions that are, as nearly as possible, identical. The test parameters include fuel type, cladding cold-work, smear density, linear power rate, operating temperature, fuel restraint, and burnup. The fuel used in the fabrication of these elements will be 95% dense, single-phase $(U_{0.8}Pu_{0.2})C$ or $(U_{0.8}Pu_{0.2})N$. The carbide fuel will be fabricated at LASL from material synthesized using the arc-melting process. Material prepared by the carbothermio reduction process may also be used. The nitride

fuel for the initial loadings will be supplied by Battelle Memorial Institute and will be prepared using the hydridenitride process. BMI has produced 15 batches of fuel of which 4 batches have been received. The cladding tubing will be Type 316 stainless steel 0.310 in. o.d. with a wall thickness of 0.012 in. The cladding was purchased through HEDL from Superior Tube Co. Both solution annealed and 20% cold worked tubing were purchased. The tubing complies, as nearly as possible, to the fuel element cladding specification RDT-E13~8. End plug material has been received from HEDL. The 20% cold-worked, Type 316 stainless steel for the end plugs is a portion of the batch of bar stock fabricated for the FFTF control rod structural components and complies to RDT-M-7-23T. Hold-down springs will be made from 0.031-in.-diam, Type 302 stainless steel, FFTF driver fuel element spacer wire material.

Shroud tubes with perforated slots are planned for use as a fuel fragment restraint mechanism in about 10 of the Series K-4 fuel elements. The actual number of elements that will use shrouds will depend on the analysis of the results for the sodium-bonded elements from the encapsulated experiments currently being examined. The suggested shroud tube geometry was selected after considering several wire restraint systems, tubular restraint systems, and packing restraint systems. The suggested shroud tube is:

TABLE 463-VIII

SERIES US100 CARBIDE EXPERIMENTS

Experiment <u>No.</u>	Fuel Type ^a	Fuel Density, <u>% T.D.</u>	Clad Material ^b	Shroud	Diametral Gap, in.	Max. Fuel Temp. at Startup, C	Max. Linear Power, Kw/ft	Status
U241	MC	93	304SS	None	0.015	1175	35.8	At EBR-II Awaiting Insertion into X136
U242	MC	93	304SS	None	0,015	1175	33.8	At EBR-11 Awaiting Insertion into X156
U243	MC	93	304SS	None	0.030	1150	33.8	At EBR-II Awaiting insertion into X156
U244	MC	93	304SS	None	0.015	1175	35.8	At EBR-II Awaiting insertion into X156
U245	MC	93	3045S	None	0.030	1150	33.8	At EBR-II Awaiting insertion into X156
U246	MC	93	316SS	None	0.015	1190	36.4	At EBR-11 Awaiting insertion into N156
U247	MC	93	316SS	None	0,030	1150	33.8	At EBR-II Awaiting Insertion into X156
U248	MC	93	31688	None	0.030	1140	36.4	At EBR-11 Awaiting insertion into X156
U249	MC	93	INC-800	Ncne	0.015	1210	36.4	At EBR-II Awaiting insertion into X156
U250	MC	93	INC-800	None	0.030	1145	36.4	At EBR-11 Awaiting insertion into X156
U251	MC	93	304SS	None	0.030	1145	36.4	At EBR-11 Awaiting insertion into X156
U252	MC	93	304SS	Vanadium-slots	0.030	1140	33.4	At EBR-11 Awaiting insertion into N156
U253	MC	93	304SS	Iron-slots	0,030	1145	33.8	At EBR-II Awaiting Insertion Into X156
U254	MC	93	304SS	3045S-slots	0.030	1140	33.8	At EBR-11 Awaiting insertion into X155
U255	МС	93	304SS	304SS-holes	0.030			At EBR-11 – Rejected Winclusion in bottom weld
U256	мс + 10 v/о м ₂ с ₃	98	304SS	Vanadium-slots	0.030	1140	34.0	At EBR-H Awaiting insertion into X156
U257	MC + 10 v/o M ₂ C ₃	98	INC-800	Tantalum-slots	0.030	1135	33.5	At EBR-11 Awaiting insertion into X156
U258	MC + 10 v/o M ₂ C ₃	98	304SS	304SS-slots	0.030	1145	33.5	At EBR-11 Awaiting insertion into X156
U259	MC + 10 v/o M ₂ C ₃	98	INC-800	304SS-slots	0.030	1150	34.6	At EBR-II Awaiting insertion into X156

- 1. perforated with slots,
- 2. slot axial pitch, 0.125 in.,
- 3. slot length, 0.100 in.,
- 4. slot width, 0.020 in.,
- 5. twenty-four slots per tube revolution,
- adjacent circumferential rows of slots staggered axially by 1/3 the axial pitch,
- circumferential ligament between slots, 0.015 in.,
- 8. axial ligament between slots, 0.025 in.,
- 9. tube dimensions, 0.2675 in. x 0.2615 in.,
- 10. slot area, 43.8%,
- 11. tube material, 25% cold-worked Type 316 stainless steel,

12. method of slot fabrication, Electro Discharge Machine (EDM).

In addition to the experiments described above, two nitride fueled thermal irradiation experiments from ORNL (43N1 and 43N2) and eight carbide fueled EBR-II experiments from Westinghouse will be examined. Results and status will be reported in future reports pending receipt of experiment description and history.

Several pieces of equipment for the fuel element loading facility are undergoing testing and modifications. The design and assembly of the xenon tagging apparatus has been completed and is ready for testing. The eddy current testing apparatus has been operated successfully

TABLE 463-IX

SERIES C-5 EBR-II NITRIDE EXPERIMENTS^a

E	xperiment	Fuel Density,	Smear Density,	Max. Fuel Temp. at	Max. Clad	Max. Lincar	Status
-	No.	<u>% T.D.</u>	<u>% T.D.</u>	Startup, F (C)	<u>Temp., F(C)</u>	Power, Kw/It	Status
	C-5-1	92.6	79.4				Reject; chips, air in plenum
	C-5-2	93.3	79.4				Reject: chips, air in plenum
	C-5-3	94.0	79.9				Reject; chips, air in plenum
	C5-4	95.1	80.9	2112 (1156)	1326 (719)	33.2	At LASL for NDT
·	C-3-5	95.4	81.6		SPARE		At LASL for NDT
	C5-6	93.3	79.3	2116 (1158)	1328 (720)	33.3	At LASL for NDT
	C-5-7	94.4	80.5	2164 (1184)	1336 (724)	33.7	At LASL for NDT
	C-5-8	94.3	74.4	2061 (1127)	1154 (623)	32.7	At LASL for NDT
	C-5-9	94.2	80.2	2071 (1133)	1203 (651)	33.5	At LASL for NDT
	C-5-10	94.1	80.2	2021 (1105)	1157 (1:25)	32.6	At LASL for NDT
	C-5-11	93.9	80.6	2050 (1121)	1165 (629)	33.4	At LASL for NDT
	C-5-12	. 94.2	74.1	2050 (1121)	1191 (644)	32.5	At LASL for NDT
	C-5-13	95.5	75.2	2036 (1113)	1154 (623)	32.1	At LASL for NDT
	C-5-14	95.7	75.4	2087 (1142)	1312 (711)	32.0	At LASL for NDT
	C-3-15	95.2	75.4	2036 (1113)	1188 (642)	32.1	At LASL for NDT
	C-5-16	95.6	75.4		SPARE		At LASL for NDT
	C-5-17	95.7	75.5			_	Reject: leaked, air in plenum
	C-5-18	93.9	79.6	2039 (1115)	1193 (645)	32.6	At LASL for NDT
	C-5-19	93.6	78.9	2054 (1123)	1200 (649)	33.0	At LASL for NDT
	C-5-20	94.7	80.6	2024 (1107)	1155 (£24)	32.4	At LASL for NDT

^aAll elements are sodium bonded. The fuel is (U_{0.8}Pu_{0.2})N pellets. The cladding is 20% cold-worked Type 316 stainless steel, 0.310 in. o.d. by 0.280 in. i.d.

TABLE 463-X

SERIES O-N1 EBR-II NITRIDE EXPERIMENTS^a

Experiment	Fuel Density, <u>% T.D.</u>	Smear Density <u>% T.D.</u>	Temp. at <u>Startup, F (°C)</u>	Max. Clad Temp., F (°C)	Max. Linear Power, Kw/ft	Status
O-N1-1	89.9	77.7	2085 (1140)	1200 (649)	32.9	At LASL for NDT
O-N1-2	90.3	78.0	2043 (1117)	1325 (718)	33.1	At LASL for NDT $^{\mathrm{b}}$
O-N1-3	90.0	77.8	2128 (1164)	1323 (717)	32.6	At LASL for NDT
O-N1-4	89.6	77.4	2133 (1167)	1323 (717)	32.8	At LASL for NDT
O-N1-5	90.4	78.1	2068 (1131)	1160 (627)	32.9	At LASL for NDT ^b
O-N1-6	89.3	77.2		SPARE		At LASL for NDT ^b
O-N1-8	89.6	77.4		SPARE		At LASL for NDT

^aAll elements are sodium bonded. The fuel is (U_{0.8}Pu_{0.2})N pellets. The cladding is 20% cold-worked. Type 316 stainless steel, 0.310 in. o.d. by 0.280 in. i.d.

^bElements O-N1-2, -5, and -6 have large fuel chips in the pellet-cladding annulus.

TABLE 463-XI

SERIES	K4	CARBIDE	EXPERIMENTS
			•

52 C 82 SA Yes 1060 (570) 1900 (1040) 7 53 C 82 SA Yes 1170 (630) 2000 (1100) 11 54 C 82 SA No 1170 (630) 2000 (1100) 11 55 C 82 SA No 1060 (570) 1900 (1040) 7 56 C 82 SA No 1060 (655) 2000 (1100) 11 57 C 82 CW Yes 1050 (565) 1900 (1040) 11 58 C 82 CW Yes 1160 (625) 2000 (1100) 11 59 C 82 CW No 1050 (565) 1900 (1040) 11 60 C 85 SA No 1050 (565) 1900 (1040) 11 62 N 82 SA No 1050 (565) 2400 (1315) 11 63 N 82 CW	Experiment	Fuel Type	Sinear Density, %	<u>Cladding</u> ^C	Shroud	Peak Cladding Temp., F (°C)	Approximate Peak Centerline Temp., F ([°] C)	Goal Burnup, at.%
53 C 82 SA Yes 1170 (630) 2000 (1100) 11 54 C 82 SA No 1170 (630) 2000 (1100) 11 55 C 82 SA No 1060 (570) 1900 (1040) 7 56 C 82 SA Yes 1160 (625) 2000 (1100) 11 57 C 82 CW Yes 1050 (565) 1900 (1040) 11 58 C 82 CW Yes 1160 (625) 2000 (1100) 11 59 C 82 CW No 1050 (565) 1900 (1040) 11 60 C 85 SA No 1050 (565) 1900 (1040) 11 62 N 82 CW No 1050 (565) 2400 (1315) 7 63 N 82 CW No 1170 (630) 2500 (1375) 11 64 N 82 CW	5 2	С	82	SA	Yes	1060 (570)	1900 (1040)	7
54 C 82 SA No 1170 (630) 2000 (1100) 11 55 C 82 SA No 1060 (570) 1900 (1040) 7 56 C 82 SA Yes 1160 (625) 2000 (1100) 11 57 C 82 CW Yes 1050 (565) 1900 (1040) 11 58 C 82 CW Yes 1160 (625) 2000 (1100) 11 59 C 82 CW Yes 1160 (625) 2000 (1040) 11 60 C 82 CW No 1050 (565) 1900 (1040) 11 61 C 85 SA No 1060 (570) 1900 (1040) 11 62 N 82 CW No 1050 (565) 2400 (1315) 11 63 N 82 CW Yes 1170 (630) 2500 (1375) 11 64 N 82 CW	53	С	82	SA	Yes	1170 (630)	2000 (1100)	11
55 C 82 SA No 1060 (570) 1900 (1040) 7 56 C 82 SA Yes 1160 (625) 2000 (1100) 11 57 C 82 CW Yes 1050 (565) 1900 (1040) 11 58 C 82 CW Yes 1160 (625) 2000 (1100) 11 59 C 82 CW No 1050 (565) 1900 (1040) 11 60 C 85 SA No 1060 (570) 1900 (1040) 11 61 C 85 CW No 1060 (570) 1900 (1040) 11 62 N 82 SA No 1050 (565) 2400 (1315) 11 63 N 82 CW No 1170 (630) 2500 (1375) 11 64 N 82 CW Yes 1180 (640) 2500 (1315) 11 65 N 82 CW	54	С	82	SA	No	1170 (630)	2000 (1100)	11
56 C 82 SA Yes 1160 (625) 2000 (1100) 11 57 C 82 CW Yes 1050 (565) 1900 (1040) 11 58 C 82 CW Yes 1160 (625) 2000 (1100) 11 59 C 82 CW No 1050 (565) 1900 (1040) 11 60 C 85 SA No 1050 (565) 1900 (1040) 11 61 C 85 SA No 1050 (565) 1900 (1040) 11 62 N 82 SA No 1050 (565) 2400 (1315) 7 63 N 82 CW No 1170 (630) 2500 (1375) 11 64 N 82 CW Yes 1180 (640) 2500 (1315) 11 65 N 82 CW Yes 1080 (580) 2400 (1315) 11 66 N 82 SA	55	С	82	SA	No	1060 (570)	1900 (1040)	7
57 C 82 CW Yes 1050 (565) 1900 (1040) 11 58 C 82 CW Yes 1160 (625) 2000 (1100) 11 59 C 82 CW No 1050 (565) 1900 (1040) 11 60 C 85 SA No 1050 (565) 1900 (1040) 7 61 C 85 CW No 1060 (570) 1900 (1040) 11 62 N 82 SA No 1050 (565) 2400 (1315) 7 63 N 82 CW No 1170 (630) 2500 (1375) 11 64 N 82 CW Yes 1170 (630) 2500 (1375) 11 65 N 82 CW Yes 1170 (630) 2500 (1375) 11 66 N 82 CW Yes 1070 (575) 2400 (1315) 11 67 N 82 SA	56	С	82	SA	Yes	1160 (625)	2000 (1100)	11
58 C 82 CW Yes 1160 (625) 2000 (1100) 11 59 C 82 CW No 1050 (565) 1900 (1040) 11 60 C 85 SA No 1050 (565) 1900 (1040) 7 61 C 85 CW No 1060 (570) 1900 (1040) 11 62 N 82 CW No 1050 (565) 2400 (1315) 7 63 N 82 CW No 1170 (630) 2500 (1375) 11 64 N 82 CW Yes 1180 (640) 2500 (1375) 7 66 N 82 CW Yes 1070 (575) 2400 (1315) 11 67 N 82 CW Yes 1080 (580) 2400 (1315) 11 67 N 82 SA Yes 1080 (580) 2400 (1315) 11 68 N 82 SA	57	С	82	CW	Yes	1050 (565)	1900 (1040)	11
59 C 82 CW No 1050 (565) 1900 (1040) 11 60 C 85 SA No 1050 (565) 1900 (1040) 7 61 C 85 CW No 1060 (570) 1900 (1040) 11 62 N 82 SA No 1050 (565) 2400 (1315) 7 63 N 82 SA No 1170 (630) 2500 (1375) 11 64 N 82 CW No 1180 (640) 2500 (1375) 11 65 N 82 CW Yes 1180 (640) 2500 (1375) 11 66 N 82 CW Yes 1070 (575) 2400 (1315) 11 67 N 82 SA Yes 1080 (580) 2400 (1315) 11 68 N 82 SA Yes 1050 (565) 2400 (1315) 11 69 N 85 SA	58	С	82	CW	Yes	1160 (625)	2000 (1100)	11
60 C 85 SA No 1050 (565) 1900 (1040) 7 61 C 85 CW No 1060 (570) 1900 (1040) 11 62 N 82 SA No 1050 (565) 2400 (1315) 7 63 N 82 SA No 1170 (630) 2500 (1375) 11 64 N 82 SA Yes 1170 (630) 2500 (1375) 11 65 N 82 CW Yes 1180 (640) 2500 (1375) 7 66 N 82 CW Yes 1070 (575) 2400 (1315) 11 67 N 82 CW Yes 1070 (575) 2400 (1315) 11 68 N 82 SA Yes 1080 (580) 2400 (1315) 11 68 N 82 SA Yes 1050 (565) 2400 (1315) 11 69 N 85 SA	59	С	82	CW	No	1050 (565)	1900 (1040)	11
61C85CWNo1060 (570)1900 (1040)1162N82SANo1050 (565)2400 (1315)763N82CWNo1170 (630)2500 (1375)1164N82SAYes1170 (630)2500 (1375)1165N82CWYes1180 (640)2500 (1375)766N82CWYes1070 (575)2400 (1315)1167N82SAYes1080 (580)2400 (1315)1168N82SAYes1050 (565)2400 (1315)1169N85SANo1070 (570)2400 (1315)1170N85CWNo1050 (565)2400 (1315)7	60	С	85	SA	No	1050 (565)	1900 (1040)	7
62 N 82 SA No 1050 (565) 2400 (1315) 7 63 N 82 CW No 1170 (630) 2500 (1375) 11 64 N 82 SA Yes 1170 (630) 2500 (1375) 11 65 N 82 SA Yes 1170 (630) 2500 (1375) 11 65 N 82 CW Yes 1180 (640) 2500 (1375) 7 66 N 82 CW Yes 1070 (575) 2400 (1315) 11 67 N 82 SA Yes 1080 (580) 2400 (1315) 11 68 N 82 SA Yes 1050 (565) 2400 (1315) 11 69 N 85 SA No 1070 (570) 2400 (1315) 11 70 N 85 CW No 1050 (565) 2400 (1315) 7	61	С	85	CW	No	1060 (570)	1900 (1040)	11
63 N 82 CW No 1170 (630) 2500 (1375) 11 64 N 82 SA Yes 1170 (630) 2500 (1375) 11 65 N 82 CW Yes 1180 (640) 2500 (1375) 7 66 N 82 CW Yes 1070 (575) 2400 (1315) 11 67 N 82 SA Yes 1080 (580) 2400 (1315) 11 68 N 82 SA Yes 1050 (565) 2400 (1315) 11 69 N 85 SA No 1070 (570) 2400 (1315) 11 70 N 85 CW No 1050 (565) 2400 (1315) 7	62	N	82	SA	No	1050 (565)	2400 (1315)	7
64 N 82 SA Yes 1170 (630) 2500 (1375) 11 65 N 82 CW Yes 1180 (640) 2500 (1375) 7 66 N 82 CW Yes 1070 (575) 2400 (1315) 11 67 N 82 SA Yes 1080 (580) 2400 (1315) 11 68 N 82 SA Yes 1050 (565) 2400 (1315) 11 69 N 85 SA No 1070 (570) 2400 (1315) 11 70 N 85 CW No 1050 (565) 2400 (1315) 7	63	N	82	CW	No	1170 (630)	2500 (1375)	11
65 N 82 CW Yes 1180 (640) 2500 (1375) 7 66 N 82 CW Yes 1070 (575) 2400 (1315) 11 67 N 82 SA Yes 1080 (580) 2400 (1315) 11 68 N 82 SA Yes 1050 (565) 2400 (1315) 11 69 N 85 SA No 1070 (570) 2400 (1315) 11 70 N 85 CW No 1050 (565) 2400 (1315) 7	64	N	82	SA	Yes	1170 (630)	2500 (1375)	11
66 N 82 CW Yes 1070 (575) 2400 (1315) 11 67 N 82 SA Yes 1080 (580) 2400 (1315) 11 68 N 82 SA Yes 1050 (565) 2400 (1315) 11 69 N 85 SA No 1070 (570) 2400 (1315) 11 70 N 85 CW No 1050 (565) 2400 (1315) 7	65	N	82	CW	Yes	1180 (640)	2500 (1375)	7
67 N 82 SA Yes 1080 (580) 2400 (1315) 11 68 N 82 SA Yes 1050 (565) 2400 (1315) 11 69 N 85 SA No 1070 (570) 2400 (1315) 11 70 N 85 CW No 1050 (565) 2400 (1315) 7	66	N	82	CW	Yes	1070 (575)	2400 (1315)	11
68 N 82 SA Yes 1050 (565) 2400 (1315) 11 69 N 85 SA No 1070 (570) 2400 (1315) 11 70 N 85 CW No 1050 (565) 2400 (1315) 7	67	N	82	SA	Yes	1080 (580)	2400 (1315)	11
69 N 85 SA No 1070 (570) 2400 (1315) 11 70 N 85 CW No 1050 (565) 2400 (1315) 7	68	N	82	SA	Yes	1050 (565)	2400 (1315)	11
70 N 85 CW No 1050 (565) 2400 (1315) 7	69	N	85	SA	No	1070 (570)	2400 (1315)	11
	70	N	85	CW	No	1050 (565)	2400 (1315)	7

^aHeating rates will be in the range 38 to 40 Kw/ft. All elements sodium bonded.

^bC = 95% dense $(U_{0.8}^{Pu}Pu_{0.2})C$, 93% enriched ²³⁵U.

N = 95% dense $(U_{0.8}^{Pu}Pu_{0.2})$ N, 93% enriched ²³⁵U.

^cSA = solution annealed Type 316 stainless steel. CW = 20% cold worked Type 316 stainless steel. Both types are 0.310 in. o.d. by 0.286 in. i.d.

with the eddy current standard and a standard operating test procedure is being prepared.

C. TREAT Irradiation Testing

(J. F. Kerrisk, D. G. Clifton, R. E. Alcouffe)

In order to assess the behavior of (U, Pu)C and (U, Pu)N fueled elements under fast reactor accident conditions, transient irradiations will be conducted in the TREAT facility. Investigations will be carried out on both irradiated and unirradiated fuel pins to determine (1) the threshold power levels at which damage or failure occurs, (2) the effect of bond and cladding defects, and (3) the failure propagation mechanism in multipin assemblies. 1. Series UL Tests. A cooperative effort has been carried out with Gulf United Nuclear Fuels Corporation in the area of TREAT testing. A series of four tests, designated LASL Series UL, will determine the effect of irradiation on the behavior of helium and sodium bonded advanced fuel elements (fabricated by GUNFC) under possible LMFBR accident conditions. Table 463-XII describes the fuel elements and test objectives. LASL has assumed complete responsibility for these tests in fiscal year 1973.

a. Tests LASL-UL-1 and LASL-UL-2. During this quarter the capsules for these tests were received

TABLE 463-XII

LASL SERIES UL TESTS

		TES	sr	
	LASL-UL-1	LASL-UL-2	LASL-UL-3	LASL-UL-4
Fucl Element	263 (138 A)	264 (146 A)	265 (138)	268 (146)
Fuel Material ^b	90 vol% (U _{0.}	85 ^{Pu} 0, 15 ^{)C + 1}	10 vol‰ (U _{0.85} I	Pu _{0,15}) ₂ C ₃
Fuel Pollet o.d., in.	0.246	0.240	0.246	0.240
Bond Material	Не	Na	He	Na
Bond Thickness (Radial), in.	0.005	0.015	0.005	0.015
Clad Material	316SS	304SS	316SS	304SS
Clad Thickness, in.	0.022	0.015	0.022	0.015
Smear Density, % Theoretical	90	77	90	77
Fuel Column Length. in.		13,75 ± 0	.125	
Burnup, MWD/MTM ^C	0	0	45,000	45,000
Test Objective	Fuel Melting	Fuel Melting	Same Tran- sient as 263	Same Tran- sient as 264

^aFuol element numbers reassigned by GUNFC. Old numbers shown in parentheses.

^bUranium enriched to 60% in 235 U.

^oIrradiated in EBR-11 at 10 to 15 Kw/ft in subassembly X-055.

from GUNFC, the two leaking TREAT capsule heads were replaced with new heads, and the capsules were inspected and prepared for shipment to TREAT. The following sequence of operations was performed in achieving these results:

1. The TREAT capsules received from GUNFC were disassembled.

2. The inner capsules were x-radiographed and thermocouple circuits were tested for electrical continuity.

3. Three TREAT capsule heads with new potted epoxy electrical seals were prepared, helium leak tested, hydrostatically tested to 15,000 psi, and helium leak tested again. A leak was found in the potted electrical seal of one of the heads after the hydrostatic test.

4. The two TREAT capsules were hydrostatically tested to 15,000 psi with no indication of leaks or damage.

5. The inner capsules and TREAT capsules were assembled using the leak tight heads and subjected to a helium leak test, an electrical continuity test, a visual examination, and a trial heat-up to pre-test temperatures using the capsule heaters.

6. Both TREAT capsules were x-radiographed in the assembled state as the last inspection prior to shipment, and loaded into their shipping container.

A quality assurance plan for tests LASL-UL-1 and LASL-UL-2 was written and accepted by the TREAT staff. This plan covers the inspections and tests performed at LASL and at TREAT prior to test in the TREAT Reactor as well as the post-test examination. Quality assurance procedures covering these inspections were also written. With the approval of this QA plan, LASL received permission to ship the capsules for these tests to TREAT.

b. Tests LASL-UL-3 and LASL-UL-4. The assembly of the inner capsules for tests 3 and 4 of the UL Series was done by GUNFC up to the point of insertion of the fuel elements into the inner capsules. Since the fuel elements were pre-irradiated in EBR-II, the completion of the assembly of the inner capsules and the assembly of the inner capsules into the TREAT capsules requires hot cell facilities. LASL, in conjunction with RDT, has arranged for the completion of the assembly by the Radiometallurgy Group of Hanford Engineering Development Laboratory. LASL has received copies of HEDL procedures covering similar assembly operations for their TREAT experiments, reviewed them, and is discussing possible changes with HEDL. The fuel elements and partially assembled inner capsules for these tests have been shipped to HEDL. A QA plan covering the completion of the assembly and inspections at HEDL and the post-test examination is in preparation.

c. Neutronic Calculations. Calculations for the UL Series experiments had been previously performed^{3,4,5} using a preliminary cross-section set consisting of 24 epithermal and 5 thermal groups. However, this set underestimates the worth of some other experiments by 0.5 to 1.0 $\Delta k/k$. Therefore, this quarter has been devoted to obtaining and testing a new cross-section set. Because of the extreme changes in spectrum through the experimental materials it was felt that 5 thermal groups were insufficient; therefore, 20 thermal groups are used in the new set. Also, the MC² code was used to obtain the epithermal cross sections from ENDF/B leading to a more accurate determination of the resonance self shielding effects. Thus the new set consists of 44 groups, 24 epithermal and 20 thermal.

A series of both one-dimensional and two-dimensional calculations were performed for the UL series of experiments to determine the changes predicted by the new set of cross sections. The results are presented in Table 463-XIII. Because the calculated reactivity worth of the experiment is closer to the expected $3\% \Delta k/k$, it appears that the use of the new set does lead to significantly improved calculations of the system.

2. Series 1 Tests. A group of eight tests using LASL fabricated fuel elements has been designated LASL Series 1 tests. The tests are designed to determine if any significant safety related behavioral problems exist for sodium bonded, stainless steel clad (U, Pu)C and (U, Pu)N fuels by defining failure thresholds and the types of failure experienced by these fuels. Table 463-XIV summarizes the test parameters and objectives. Approval-in-principle has been received from the AEC for this series of tests. Design work for the modification of TREAT capsules to incorporate a thermal neutron filter and calculations for a preliminary safety analysis are in progress.

Neutronic calculations have also been performed for the Series 1 experiments using the new cross-section set, and based on a 0.010-in.-thick gadolinium thermal neutron filter and a preliminary capsule design. Table 463-XIII shows the results obtained.

TABLE 463-XIII

RESULTS OF NEUTHONIC CALCULATIONS

	SERIES UL		SER	ES 1
	Current Results	Previous Results	Current Results	Previous Results
One Dimensional Transport Calculation				
Edge/Center Power Generation Ratio	6.08	7.53	1.91	2.27
Figure of Morn ^b	1.211x10 ⁻⁴	1.083x10 ⁻⁴	0.459x10 ⁻⁴	0.397x10 ⁻⁴
Experiment Worth (Ak/k)	2.87%	1.90%	5.2%	4.6%
Two Dimensional Generalized				
Diffusion Equation Calculation				
Figure of Merit ^b	0.917×10 ⁻⁴	1.170×10-4	0.224x10 ⁻⁴	-

²Sce References 3-5.

 $^{b}\mathrm{F}$ truer of merit is the ratio of the power density in the fuel to the power produced by the reactor in W/cc/W.

^CSee Reference 6.

TABLE 463-XIV

LASL SERIES 1 EXPERIMENTS

<u>Test</u>	Fuel Material ^a	Burnup	TREAT Transient	Test Objective
1A-1	(U _{0.8} Pu _{0.2})C	0	Fast	Na bond ejection- incipient fuel melting
1A-2	(U _{0.8} ^{Pu} 0.2)C	0	Slow	Same as 1A-1
1B-1	(U _{0.8} ^{Pu} 0.2)C	0	Fast	50% Fuel melting
1B-2	(U _{0.8} ^{Pu} 0.2)N	0	Fast	Same as 1B-1 ^d
1B-3	(U _{0.8} ^{Pu} 0.2)C	Ũ	Slow	Same as 1B-1
1B-4	(U _{0.8} Pu _{0.2})N	0	Slow	Same as 1B-1 ^d
1C-1	(U _{0.8} ^{Pu} 0.2)C	8%	c	Same as 1B-1
1C-2	(U _{0.8} ^{Pu} 0.2)N	8%	c	Same as 1B-1 ^d

⁸The fuel will be pellets, contained in 0.310 in. o.d. by 0.012-in. wall 316 stainless steel eladding at 80% smear density. The uranium is enriched to 93% in 235 U.

^bFast transients will deposit energy in thue periods of the order of 1 see while slow transients will be on the order of 10 see.

^cThe type of transient to be used will be determined by the results of the unirradiated element tests.

^dSince (U, Pu)N does not melt, but decomposes to metal and nitrogen. the test objectives are described in terms of the energy required to produce a given melting in (U, Pu)C.

III. FUEL PROPERTY MEASUREMENTS

Differential Thermal Analysis (J. G. Reavis)

The program of differential thermal analysis of irradiated UO_2 -PuO₂ fuel materials supplied by G.E.-Sunnyvale is continuing. A sample of UO_2 -25% PuO₂ irradiated to 7.6 at.% burnup has been sealed in a W capsule for observation. A Ta shim was incorporated into the weld in an attempt to improve the welding characteristics and increase the probability of achieving a crack-free weld.

Before the capsule of irradiated fuel could be observed, however, problems appeared in the vacuum system. Portions of the system exterior to the cell have been reconditioned, but leaks apparently still exist inside the hot cell.

IV. REFERENCES

- R. D. Baker, 'Quarterly Status Report on the Advanced Plutonium Fuels Program, April 1 to June 30, 1972 and Sixth Annual Report, FY 1972," Los Alamos Scientific Laboratory Report LA-5067-PR (1972).
- D. L. Keller, "Progress on Development of Fuels and Technology for Advanced Reactors During July, 1971 Through June, 1972," Battelle Memorial Institute Report BMI-1925 (1972).
- J. C. Vigil, "Theoretical Reactor Physics Program and First Annual Report, FY 1971," Los Alamos Scientífic Laboratory Report LA-4738-MS (1971).
- J. C. Vigil, "Theoretical Reactor Physics Program, October 1-December 3, 1971," Los Alamos Scientific Laboratory Report LA-4889-PR (1972).
- J. C. Vigil, "Theoretical Reactor Physics Program, January 1-March 31, 1972," Los Alamos Scientific Laboratory Report LA-4955-PR (1972).
- R. E. Alcouffe, "A Generalized Finite Differential Diffusion Equation for Neutron Transport Computations." Los Alamos Scientific Laboratory Report LA-4938-MS (1972).

CORRECTION - PROJECT 463

In LA-5067-PR, 'Quarterly Progress Report on the Advanced Plutonium Fuels Program, April 1 to June 30, 1972" the figures on pages 35 and 36 are mixed up: Fig. 463-2 -- Title is correct, but graph shown (p. 35) as Figure 463-3 should be here. Fig. 463-3 -- Title is correct, but graph shown (p. 35) as Figure 463-4 (or a 20) should

(p. 35) as Figure 463-4 (on p. 36) should be here.

Fig. 463-4 -- Title is correct, but the graph (p. 36) shown as Fig. 463-2 (p. 35) should be here.

PROJECT 472

ANALYTICAL STANDARDS FOR FAST BREEDER REACTOR OXIDE FUEL

Person in Charge: R. D. Baker Principal Investigator: G. R. Waterbury

I. INTRODUCTION

Necessary to the development of the high quality fuels and control rods required by the LMFBR program are highly reliable analytical methods for the chemical characterization of the source materials and the pellet products, and for the measurement of burnup and fission gases on irradiated fuels. Objectives concerned with ensuring the production of these high quality reactor materials are: (1) the continued preparation of carefully characterized calibration materials for the various analytical methods; (2) the preparation of quality control samples used for surveillance of analytical chemistry laboratory operations during periods of production; (3) continued assistance and guidance of quality assurance programs for chemical specifications sampling and analysis; (4) the development of more reliable methods of chemical analysis with emphasis on O/M measurement, (5) the preparation of continuously updated compilations of analytical methods for these materials; (6) the analysis, as a reference laboratory, of those samples in dispute between a vendor and a purchaser. For the immediate future, these objectives are centered on the FFTF. Later, they will be extended to the LMFBR demonstration and large production plants.

Objectives concerned with irradiated LMFBR fuel examination are: (1) the development of burnup methods based on conventional mass spectrometry measurements and later by spark source mass spectrometry; (2) the development of faster burnup methods using chemical analysis techniques; and (3) start development of analytical methods for the determination of gases in pre and postirradiated fuels and control rod absorber materials to provide data on gas retention properties and cladding stability.

II. ANALYTICAL CHEMISTRY PROGRAM FOR BORON CARBIDE

A program equivalent to that established for the production of FFTF mixed oxide fuel is being set up for the production of FFTF boron carbide pellets.

A. Status of Analytical Methods and Qualification of Analytical Laboratories (R. K. Zeigler, J. E. Rein, G. R. Waterbury)

Last year, two AEC-Internal Round Robins indicated that analytical methods for the measurement of nine chemical properties of boron carbide were satisfactory relative to the requirements set by HEDL for the FFTF. Copies of these methods were distributed to the three potential vendors of FFTF boron carbide pellets and a round robin with vendor participation plus two AEC laboratories (HEDL and LASL) was conducted. Only the results for total carbon and boron isotopic distribution were considered adequate for specification analyses. Either the betweenlaboratory difference or the measurement reproducibility were too large for the other seven (total boron, HCl soluble boron, HNO₃ soluble boron, soluble carbon, chloride, fluoride, and general metallic impurities) methods.

The three vendor laboratories were visited by HEDL and LASL personnel to provide technical aid with emphasis on those methods where the round robin results indicated difficulties. It is believed that the discussions disclosed many likely sources of difficulty. Characterized boron carbide batches were given to the laboratories to be used for the evaluation of method modifications. A second vendor round robin with the same three industrial laboratories and two AEC laboratories has been started. The methods to be evaluated are the seven that were not deemed satisfactory in the first round robin. Completion is scheduled for early November.

в.	Preparation of Calibration Materials and Quality
	Control Samples
	(J. A. Pena, O. R. Simi, H. J. Kavanaugh,
	J. E. Rein)

The preparation of these materials, in amounts estimated to be used by the vendor laboratories and by HEDL for the production of the FFTF boron carbide pellets, is nearing completion. Characterization of the materials, necessary for the verification of homogeneity, is under way.

C. Status of RDT Standards

(J. E. Rein, R. K. Zeigler, W. H. Ashley, O. R. Simi, G. R. Waterbury)

Jointly with HEDL, a draft of RDT standard F11-2 "Analytical Chemistry Methods for Control Rod Absorber Material" has been prepared and submitted to RDT for review and approval. The analytical methods included in this document are the nine presented previously in Section A plus gas content and water content.

A draft of RDT standard F2-8 'Qualification and Control of Analytical Chemistry Laboratories for Control Rod Absorber Material Analysis, " also prepared jointly by HEDL and LASL, is essentially complete and is expected to be submitted to RDT early next quarter. This document prescribes the course of action by which analytical laboratories establish their technical competence to do the chemical analyses required for the characterization of boron carbide pellets. It also describes the quality control program that will be in effect for the continual evaluation of the analytical data that will be obtained during periods of boron carbide pellet production.

D. Studies and Improvements of Analytical Methods

 (R. D. Gardner, A. L. Henicksman, W. H. Ashley)
 1. Determination of Soluble B. Soluble B is

expected to exist in B_4C as B_2O_3 and elemental B. These minor components are determined after selective dissolution of B_2O_3 in 0.1M HCl from one sample, and both B_2O_3 and B in 1.6M HNO₃ from a second sample. As the boron carbide materials cover a composition range from about $B_{12}C_3$ to $B_{13}C_2$, the possibility of dissolving B from high boron content carbides (near $B_{13}C_2$) was investigated. This was done by refluxing material described as $B_{6.1}C$ with $1.6\underline{M}$ HNO₃ for 4 h according to the method. The dissolved B was 0.47% of the sample weight which was within the concentration range usually found. Then the residue from the first refluxing was separated and washed, and the refluxing with $1.6\underline{M}$ HNO₃ for 4 h was repeated. Only 0.03% of B was dissolved during the second refluxing, indicating that the high boron content material does not continue to dissolve significantly. These results indicate that the method probably is valid for materials having a wide range of boron-carbon compositions.

As we are comparing the soluble boron values obtained by refluxing in HNO_3 with those obtained by dissolution in H_2O_2 , the double leaching experiment described above was repeated using 10% H_2O_2 as the solvent after the first leach was 0.42% as compared to 0.04% after the second leach. The small difference between 0.47% and 0.42% is well within the limits of variation for the HNO_3 soluble B in a sample. Either method seemed to be equally satisfactory for this measurement.

2. Determination of Total B. Summations of total carbon, total boron, and the measured impurities have been consistently less than 100%. In an attempt to determine the cause for the low results, we analyzed the hydrous oxides, which are discarded in the boron procedure, and found only negligible amounts of boron. The stability of primary standard SRM 951, following exposure to air for 1 yr, was verified by analysis. The probable cause for low results was found to be the iron contamination introduced during grinding in a diamond (steel) mortar. In the latest set of total boron analyses, the iron in the fusion solutions was found to vary from 0.18% to 0.56%. This impurity interferes with the titration of the dissolved boron with 0.1N NaOH. As the iron is probably introduced in discrete particles of varying size, it may not be uniformly distributed. Although boron results could be corrected for the intereference, if each sample portion were analyzed for Fe, such a correction is not applicable for the carbon determination. The use of a different mortar, possibly boron carbide, is recommended for grinding the samples.

III. ANALYTICAL CHEMISTRY PROGRAM FOR FBR MIXED OXIDE FUEL

A. Qualification of Analytical Laboratories (R. K. Zeigler, J. E. Rein, G. R. Waterbury)

The laboratories of the two fuel vendors are in the process of becoming qualified for the analysis of the fuel and source materials to be manufactured for the FFTF. Aid is being given to HEDL, as requested, in this project. This includes providing of calibration materials, review of the data reported by the laboratories where discrepancies occur, and technical review of the analytical methods that are used in an attempt to trace causes of difficulty. B. Calibration Materials and Quality Control Samples

(J. A. Pena, H. J. Kavanaugh, C. J. Martell, W. M. Myers, C. B. Collier, J. E. Rein)

The quantities of these materials, estimated to be used for the qualification phase of two fuel vendors and for the first quarter of fuel production, were provided to HEDL in May 1972. Since that time, changes have been made in the quality assurance program that will result in an increased usage of both calibration materials and quality control samples by a factor of at least two. These changes include (1) addition of the plutonium dioxide supplier (ARHCO) to the quality assurance program, (2) increase in the production rate of fuel, (3) increased coverage from eight to nine quarters of fuel production, and (4) increases in the amount of material required for an analysis by some of the laboratories for some of the analytical methods.

The overall effect is that a very significant increase in effort will be necessary to prepare, characterize, package, and ship the materials to meet the greater needs. Materials on hand are now being packaged for shipment to HEDL and the preparation of new material batches has been started.

C. Status of RDT Standards (J. E. Rein, R. K. Zeigler, G. R. Waterbury)

RDT approved the draft of RDT standard F11-1 "Analytical Chemistry Methods for Mixed Oxide Fuel" prepared by HEDL and LASL. A draft of RDT standard F2-6 "Qualification and Control of Analytical Chemistry Laboratories for Mixed Oxide Fuel Analysis" should be completed in the near future and submitted to RDT for review and approval.

D. Development of Analytical Methods

<u>1. Determination of Burnup</u> (S. F. Marsh, J. E. Rein

The most reliable technique for the determination of nuclear fuel burnup is based on isotope dilution mass spectrometry of fission product neodymium, uranium, and plutonium using the triple spike technique. For laboratories without a mass spectrometer or those having a large routine sample load, a burnup method which requires less and lower-cost equipment is needed. Such a method would relate the number of fission events (using a selected fission product element or group of elements as the fission monitor) to the total heavy element content (based on postirradiation uranium and plutonium). The proposed scheme is a chemical separation of the selected fission monitor, uranium, and plutonium from a dissolved fuel sample followed by chemical measurements of each of the three components by a technique such as spectrophotometry.

The fission products most useful as fission monitors for FBR fuels are rare earths and elemental zirconium. Of these two, total of the rare earths is preferable because the fission yield is higher and more constant for different fissile nuclides.

A promising separation scheme being investigated is retention of the rare earth group on a cation-exchange column from ethanol-hydrochloric acid medium while U, Pu, Am-Cm, and many fission products pass through to an anion-exchange column which retains U and Pu. The rare earths would be eluted from the first column with aqueous hydrochloric acid and purified further, if necessary, prior to their chemical determination as a group. Plutonium and uranium would be sequentially eluted from the second column with $12\underline{M}$ HCl-0. $1\underline{M}$ HI and $0.1\underline{M}$ HCl, respectively, prior to their chemical determination.

The most difficult separation to achieve in the proposed scheme is that of the rare earths from the trivalent actinides Am and Cm. This separation is desirable for two reasons: to minimize the radiological hazard to personnel and to avoid the error in the rare earth determination due to trivalent actinide contributions. An investigation of the cation-exchange behavior of Eu and Am based on batch-contact experiments covering a wide range of ethanol-hydrochloric acid mixtures has indicated that nearly complete separation of the two elements can be achieved. As the equilibrium rates in ethanol-hydrochloric acid media are much slower than for aqueous media, additional studies to establish the conditions for successful column operation are needed.

2. Investigation of the Determination of O/M (G. C. Swanson)

The oxygen to heavy-metal-atom ratio (O/M) for mixed $(U, Pu)O_2$ fuels is an important parameter affecting fuel operating characteristics. The most widely used measurements of O/M are thermogravimetric in which the weight change of the sample is measured when the sample is subjected to conditions of temperature and gas composition which will produce stoichiometric (O/M = 2,000) oxide.

These conditions have been subject to much controversy: temperatures from 750° C to 1000° C and several gas mixtures are used. Standard reference materials having certified O/M ratios do not exist, and the best reference materials now available are UO₂ and PuO₂ carefully prepared by burning the highly pure metals. Changes in stoichiometries of these two oxides and their mixtures can be calculated readily from initial metal weights, but these prepared oxides differ from the solid solution mixed (U,Pu) oxides because of the expected differences in chemical potentials of mechanical mixtures and true solid solutions.

A study of thermogravimetric measurements of O/M has been started. A Mettler recording vacuum thermoanalyzer has been obtained and modified for use in a glovebox. Installation of the thermobalance is essentially complete. Several accessories for the balance, including a high vacuum system, a DTA amplifier, and an automated digital data logging system, are expected to be installed next quarter. As installation of the high vacuum system will require access to the balance interior, α -active material will not be handled in the apparatus prior to this installation.

KT:257(25)