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# Integrating Chemistry into 3D Climate Models: Detailed Kinetics in the Troposphere and Stratosphere of a Global Climate Model

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#### Abstract

This is the final report of a three-year, Laboratory Directed Research and Development (LDRD) project at Los Alamos National Laboratory (LANL). The motivation for the project is to create the first complete, threedimensional climate model that enfolds atmospheric photochemistry. The LANL chemical global climate model (GCM) not only distributes the trace greenhouse gases and modifies their concentrations within the detailed photochemical web, but also permits them to influence the radiation field and so force their own transport. Both atmospheric chemistry and fluid dynamics are nonlinear and zonally asymmetric phenomena. They can only be adequately modeled in three dimensions on the global grid. Our kineticsaugmented GCM is the only program within the atmospheric community capable of investigating interaction involving chemistry and transport. We have conducted case studies of timely three-dimensional chemistry issues. Examples include ozone production from biomass burning plumes, kinetic feedbacks in zonally asymmetric transport phenomena with month- to yearlong time scales, and volcano sulfate aerosols with respect to their potential effects on tropospheric ozone depletion.

## Background and Research Objectives

Historically within the atmospheric sciences community a segregation has existed between studies of chemistry and dynamics. This division has persisted even into the era of global climate change. A third of the greenhouse effect will likely be attributable to trace species with distributions controlled by an intimate interplay of photochemical production and loss with transport, the primary examples being ozone and methane (IPCC, 1992). Yet the chemical kinetics underlying climate have almost invariably been simulated interactively only in contexts of two dimensions or less in which atmospheric motions must

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be parameterized to a high degree (Ramanathan et al., 1987). The few chemical calculations actually performed on three-dimensional global climate model (GCM)-style grids have almost invariably occurred off line (noninteractively), so that greenhouse alterations could not feed back through physics routines into dynamics. A subsidiary segregation has been apparent within the transport regime. Atmospheric kinetics computations often deal with regions above and below the tropopause separately.

Our research project was to bridge both these gaps simultaneously by inserting detailed photochemistry into the troposphere and stratosphere of a GCM. From a computational standpoint, the project has been challenging. Full kinetic modeling of the stratosphere requires integration of more than 50 individual species reacting through over 150 photochemical channels. The stratospheric system is highly stiff, meaning that time constants for the constituents vary widely. Tropospheric chemistry is several times more complex, and the hundreds of disparate atmospheric chemical tracers and reactions must be condensed onto massive grids containing a hundred thousand or more cells. The conjunction of large numbers of chemical species and grid cells with stiffness has long kept chemistry-climate interaction at the frontiers of computational feasibility. Even the most experienced atmospheric science groups have to date been unwilling to tackle the problem. Los Alamos National Laboratory (LANL), however, has been in a unique position to succeed in construction of a chemical GCM. The Earth and Environmental Sciences (EES) Division at Los Alamos has been laying the necessary groundwork for several years. Advances that EES researchers have made now place the chemical coupling well within the reach of the atmospheric sciences.

EES researchers had run extensive tests of the Los Alamos general circulation model in anticipation of incorporating detailed kinetics. An off line chemical tracer model (CTM) attachment was developed to compare calculated and observed distributions of several greenhouse species, thus validating wind fields for dispersion of trace gases (Kao et al., 1992). More importantly, EES was the only atmospheric group in the world with real experience simulating three-dimensional (3-D) interactions of chemistry and climate. First, Kao et al. (1990) incorporated a key subset of the complete inorganic chemical scheme into the stratosphere of the GCM. In the past ten years, we have collaborated with researchers at the University of California campuses to build sophisticated kinetics packages fast enough to reproduce the entire photochemical set across hundreds of thousands of cells (Elliott et al., 1993 a and b). Preliminary but fully interactive tests within the LANL GCM stratosphere indicated that detailed photochemical-climate calculations were feasible even on older vector machines (Elliott et al., 1993b). The

kinetics are also flexible in terms of their reactions schemes, parallelized for a smooth transition to the new generation of parallel processing supercomputers, and modular so that they are compatible with other transport frameworks. Our chemical packages have ultimately been designed to function within earth system models blending physics with the major biogeochemical cycles.

# Importance to LANL's Science and Technology Base and National R&D Needs

Among the missions of the DOE is consideration of a national energy strategy. Energy usage drives climate change in all senses, and the kinetically active greenhouse gases are no exception. Tropospheric ozone production is controlled by oxidation of hydrocarbons released during fossil fuel combustion. Stratospheric ozone loss stems from CFC releases, which track industrial output. Methane increases are associated with population growth, which is energy limited. Climate research at LANL contributes to the shaping of an energy strategy. Our pioneering chemistry interactive GCM will not only compliment DOE and Laboratory efforts in energy management, but may well propel us to the cutting edge of a new and exciting field. Emphases will encompass tropospheric increases and stratospheric depletion as well as modeling to understand them.

Comprehension can only come through interactive photochemical and climate simulations. Our work has provided Los Alamos and the DOE with a sophisticated basis for analysis of all global ozone problems. Concern for the environment is another essential element of competitiveness in the coming world economy, and the impact of manufacturing on the terrestrial system can only be assessed through global modeling efforts.

At least three projects at LANL will benefit from cross fertilization with our work:

- Our kinetics packages have already demonstrated extreme efficiency on the global grid
  and, given proper interfacing, will run swiftly on the regional scale as well. LANL
  operates both the RAMS and HOTMAC mesoscale models. Support for their
  maintenance has come through DOE ASCOT and ARM funding.
- The LANL CHAMMP project studies the conversion of complex geophysical models to
  massively parallel processors. Since chemistry can be independent of neighboring
  cells, and our program architecture accounts for parallelism by maximizing vector
  lengths, the kinetics are already adapted for parallel GCMs.

• Our chemical computations can well compliment measurements resulting from the LIDAR technique, which proves efficient for studying atmospheric constituent fields.

# Scientific Approach and Accomplishments

The foundation for the photochemical experiments has been the LANL GCM (Kao et al., 1990). The program is spectral with an equivalent Eulerian grid of 4.5 degrees latitude by 4.5 degrees longitude. Vertical resolution has been enhanced to a total of twenty levels—twelve tropospheric and eight stratospheric. Radiative forcing includes long- and short-wave ozone bands. We began formulation of the complete 3-D chemistry climate program with a series of tropospheric kinetics experiments in the off line Chemical Tracer Model (CTM). Experimentations with hydrocarbon configurations have reduced uncertainties in organic reaction schemes (Elliott et al., 1996). In the troposphere, we have focused on ozone precursors—the hydrocarbons, and nitrogen and hydrogen oxides. In the stratosphere, chlorine/ozone chemistry has been investigated within disturbed wave structures. As parallelized climate models find their way onto multiple processing machines, our chemistry can readily be integrated, since it was designed precisely to make this transition. Parallel calculation approaches will someday allow modeling of the entire biogeochemical/climate system, and our modular photochemistry packages can play a vital role.

The major problems in the 3-D global photochemistry modeling in terms of computational speed, numerical instability, and chemical species mass conservation have been remedied using our improved numerics and automation with a novel chemical parameterization. A 60-species stratospheric dataset has been inserted into the LANL GCM, and coupled chemical-dynamical simulations have reached one annual cycle in length. The results show that the diurnal behaviors of short lived species are well captured even though a relatively large time step is used in the integration. The global features of ozone variation are in agreement with satellite measurements; however, with 10% overestimation by the model simulations. This discrepancy may well be due to the anthropogenical effects. In the tropospheric regime, we have been utilizing automated continuity equations within the kinetics codes to experiment with organic oxidation chains affecting ozone and methane distributions. An integrator with complete hydrocarbon kinetics developed at Turco's group at UCLA has been tested. We have also adopted NCAR CCM2 as another testbed for our photochemistry development. The NCAR CCM2

is a user-friendly, plug-compatible, and well-coded GCM that has been used at LANL since 1993. We have recently conducted a 40-year simulation (Kao et al. 1996) with realistic sea-surface temperatures observed from 1950-1989 as lower boundary conditions to drive the NCAR CCM2. The semi-Lagrangian technique for advecting positive-definite species mixing ratios is used for our chemical experiments in CCM2.

We have conducted case studies in the tropospheric CTM of timely3-D chemistry issues. Examples include simulation of ozone production from biomass burning plumes (Kao et al., 1994) and subsonic aircraft emissions. In the coupled chemical GCM stratosphere we have investigated kinetic feedbacks in zonally asymmetric transport phenomena with month- to year-long time scales. Examples include breaking waves (Kao et al., 1995), rapid poleward transport and heterogeneous chemistry. Volcano sulfate aerosols have been studied with respect to their potential effects on tropospheric ozone depletion (Zhao et al., 1996). We have incorporated the final tropospheric kinetics package into CCM2 and examined the tropospheric problems interactively. Comparisons with our chemical GCM results have calibrated lower dimensionality calculations (Zhao et al., 1996, submitted).

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