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URANIUM ALLOY DEVELOPMENT

PART VII

URANIUM ALLOYS WITH TWO ATOMIC PER CENT OF SOME RARE AND PRECIOUS METALS

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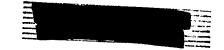
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REPORT WRITTEN BY:

A. U. Seybolt

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ABSTRACT

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The following 2 atomic per cent uranium alloys were made in 160 g ingots, and the hardness and microstructure investigated after a variety of heat treatments: V, Au, Fd, Os, Ru, Ir, Re, Rh, and Ga. All of the alloys were more or less brittle as quenched from 900° C., except those containing palladium and rhodium, and appear not to warrant further study. The rhodium alloy shows some promise of being a useful high strength uranium alloy, but further work is needed to confirm this.

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URANIUM ALLOYS WITH TWO ATOMIC PER CENT OF SOME RARE AND PRECIOUS METALS

Part VII

In the report of February 15, $1944_{4,0}$ (LA = 68) a series of 1 gram alloy melts containing 2 atomic per cent V_0 Au₀ Pd, Os, Ru, Ir, Re, Rh, and Ga were briefly described. The microstructures and hardness values in the 900° C. asquenched condition were given. Because the hardness values in general were quite high, it appeared as though at least some of these compositions might be promising for a high strength uranium alloy. Accordingly, a series of 160 g melts of 2 atomic per cent of each of the metals listed above were prepared in the usual way for further study.

Effect of Heat Treatment on the Hardness of the Alloys

All the castings were first homogenized at 900° C. for at least two hours before subsequent heat treatment and quenching at lower temperatures. The hardness value observed at each temperature level is shown in Table I.

It will be observed that a few alloys were cracked as cast, and most of them were cracked as quenched from 900° C. These groups of alloys can practically be eliminated from further consideration because of excessive brittleness. The rhenium and osmium alloys were especially brittle, and exceedingly difficult to cut. While most of the alloys show possibility of improvement in hardness on heat treatment, none shows a minimum hardness as quenched from higher temperatures, which is generally desirable for a heat treatable alloy; but the minimum hardness occurs at 700° C. or 600° C. Also, most of the alloys show a maximum hardness both at 900° C. and at 400° C.





Table I. Compositions and Rockwell A Hardness of Heat Treated 2 Atomic Per Cent Alloys All Alloys Originally Quenched from 900° C.

Alloy	Intended (Composition	Analysis Weight %	As cast	900	800	700	600	500	1
No.	Atomic%	Weight %								
2406	2▼*	0.45 V	0.465 top 0.384 bottom	59	71	71 7	60	60	69	APPR
2413	2 Au *	1.03 <u>A</u> u	1.40 top 1.18 bottom	63	65	64	56	60	69	APPROVED
24114	2 Pd	0.913 Pd	1.01 top 1.03 bottom	66	70	70	61	57	69	FOR E
24,15	2 08 +*	1.61 Os		61	66	65	63	63	61	PUBLIC
St16	2 Ru *	0.87 Ru	₩₩¢;₩\$)₩6	66	72	70	69	64	69	1
2420	2 Ir **	1.6 3 Ir	1.40 top 1.24 bottom	72	65	63	63	59	60	RELEASE
24,21	2 Ro **	1.57 Re	ఴ ౚఴౚఴౚ	73	65	65	66	65	65	۴ 🛱
2422	2 Rh	0.88 Rh	\$ ₽₽₽₩₩ ₩\$	63	73	72	70	63	69	-
2423 UN	2 Ga *	0.59 Ga	0.30 top 0.74 bottom	63	62	64	61	56	61	
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The rhodium alloy appears to show a slightly greater hardness than the palladium alloy as quenched from 900° C. and as re-heated to 400° C. Microstructure

Microstructures of all the alloys at every stage of heat treatment are shown in the following Figs., 1=63.

Vanadium Alloy, Pigs. 1-7

Most of the structures shown appear to consist of two phases, one of them sometimes rather finely divided.

Gola Alloy, Figs. 8-14

The gold alloy appears to consist of a single phase at 900° C. and possibly at 600° C. Many of the areas show cracks which had formed on the initial 900° C. heat treatment.

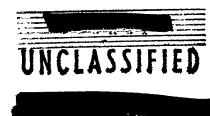
Palladium Alloy, Figs. 15-21

The alloy appears to be nearly one phase at 800° C., but at other temperatures seems to consist of two phases. The peculiar nodular structure exhibited at 700° C. is probably caused by growth of a new phase from the old structure. It is possible that the matrix of the structure in Fig. 19 consists of the nodular structure grown completely together.

Osmium Alloy, Figs. 22-28

Most of the structures observed have two phases with the possible exception of the 500° C. and 400° C. heat treated ones. The large dark patch in Fig. 28 is a crack, and all the alloys were cracked from the 900° C. condition. Ruthenium Alloy, Figs. 29-35

The as cast alloy shows a well defined two phase Muntz metal type



structure. The 600° C. heated structure seems to be almost a single phase, but there may be a fine precipitate present.

Iridium Alloy, Figs. 36-42

Here again the structures are quite varied, most of them consisting of two phases, with the 500° C. condition possibly showing a single phase structure. Rhenium Alloy, Figs. 43-49

The structures observed after quenching from 900° C., 600° C., and 500° C. are apparently all essentially single phase ones. It is possible that these structures correspond to gamma, beta, and alpha uranium, but this is only a guess. Enodium Alloy, Figs. 50-56

The as cast alloy shows a well defined Muntz metal type two phase structure. Quenching from 900° C. did not retain all of the rhodium in solid solution, and at 800° C. while it is mostly one phase, there is evidence of a finely dispersed second phase. Fig. 53 shows an interesting structure, almost certainly a non-equilibrium one, with a suggestion of three phases present.

At 600° C. there is a copious fine precipitate in a matrix apparently containing most of the rhodium, and at 500° C. the structure appears to consist of a single solid solution. At 400° C. this solution breaks down to form two phases, one of them again a finely divided one, apparently precipitated from the solid solution of Fig. 55.

Gallium Alloy, Figs. 57-63

There seems to be a small amount of a second phase in all the structures, but most of them appear to consist largely of one phase.

It must be emphasized that the micrographs shown in most cases do not correspond to equilibrium structures. While as far as analyses on these alloys

are available only the gallium showed serious sogregation, there is some possibility of segregation in the unanalyzed alloys (Os, Ru, Re) which might give misleading impressions.

The results described are only preliminary, and are intended mostly as a guide to possible future work.



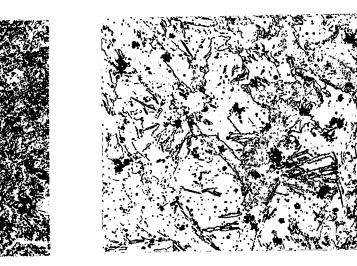






Fig. 2. 2 Atomic Por Cont V-U Alloy

Electrolytically etched in 5% Oxalic



As Quenched 900° C.

Acid

2406-2=0

2 Atomic Por Cent V-U Alloy Fig. 1. As Cast

Etched in 1:1 Nitric=Acetic Acid

2406-1-1

x 250

x 250

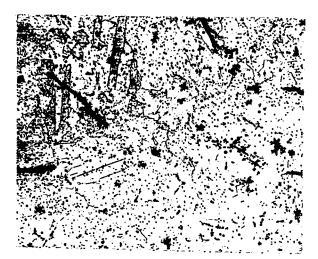
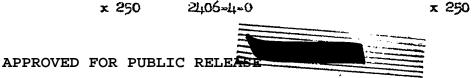


Fig. 3. 2 Atomic Per Cent V-U Alloy As Quenched 800° C.

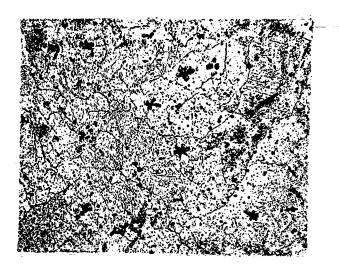
Electrolytically etched in 10% Oxalic Acid

Fig. 4. 2 Atomic Per Cent V-U Alloy As Quenched 700° C. Etched in 1:1 Nitric-Acotic Acid





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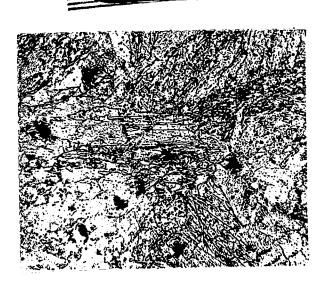


Fig. 5. 2 Atomic Per Cent V-U Alloy As Quenched 600° C. Electrolytically etched in 2% Oxalic Fig. 6. 2 Atomic Per Cent V-U Alloy As Quenched 500° C. Electrolytically etched in 2% Oxalic Acid

2406-5-0

Acid

2406-6-1

x 250

1

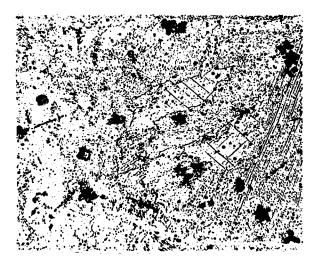
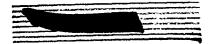


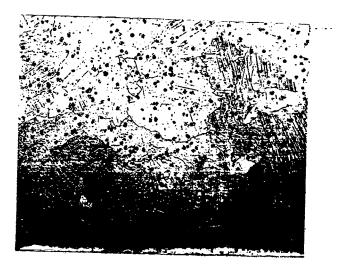
Fig. 7. 2 Atomic Per Cent V-U Alloy As Quenched 400° C. Electrolytically etched in 2% Oxalic Acid

2406-7-0

x 250



-10-



· ,.

Fig. 9. 2 Atomic Per Cent Au-U Alloy

Electrolytically etched in 2% Oxalic

As Quenched 900° C.

Fig. 8. 2 Atomic Per Cent Au-U Alloy As Cast Electrolytically etched in 2% Oxalic Acid

2413-1-0

2413-2-0

x 250

Acid

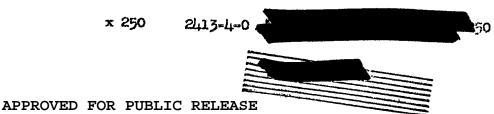
x 250



Fig. 10. 2 Atomic Per Cont Au-U Alloy As Quenched 800° C. Electrolytically etched in 2% Oxalic Acid



Fig. 11. 2 Atomic Per Cent Au-U Alloy As Quenched 700° C. Electrolytically etched in 2% Oxalic Acid





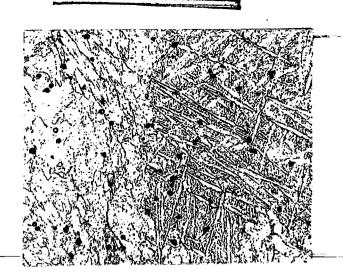


Fig. 12. 2 Atomic Per Cent Au-U Alloy As Quenched 600° C. Electrolytically stohed in 2% Oxalic Acid

Fig. 13. 2 Atomic Per Cent Au-U Alloy As Quenched 500° C. Electrolytically stohed in 2% Oxalic Acid

2413-5-0

x 250 2413-6-0

x 250



Fig. 14. 2 Atomic Per Cent Au-U Alloy As Quenched 400° C. Electrolytically etched in 5% Oxalic Acid

2413-7-0



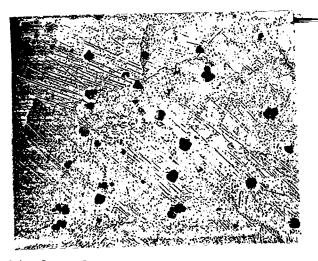




Fig.15. 2 Atomic Per Cent Pd-U Alloy As Cast

Electrolytically etched while polishing.

Fig. 16. 2 Atomic Per Cent Pd-U Alloy As Quenched 900° C. Electrolytically etched in 5% Oxalic Acid

2414-1-0

x 250

21,14-2-0

x 250

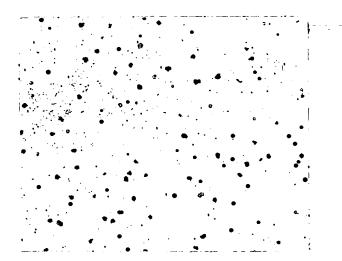


Fig. 17. 2 Atomic Per Cent Pd-U Alloy As Quenched 800° C. Etchcd in 1:1 Nitric-Acetic Acid

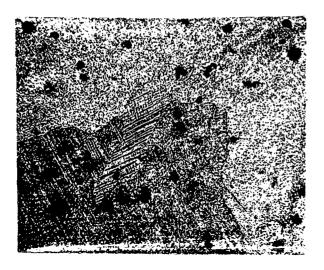


Fig. 18. 2 Atomic Per Cent Pd-U Alloy As Quenched 700° C. Electrolytically stched in 5% Oxalic Acid

24114-4-0

x 250

A 79



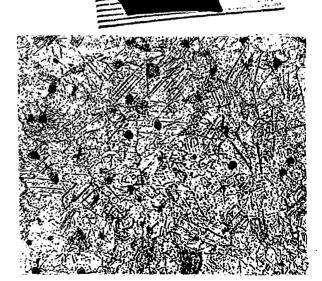


Fig. 19. 2 Atomic Per Cent Pd-U Alloy As Quenched 600° C. Electrolytically etched in 5% Oxalic Acid

Fig. 20. 2 Atomic Per Cent Pd-U Alloy As Quenched 500° C. Electrolytically etched in 5% Oxalic Acid

2414-5-0

2414-6-0 x 250

x 250

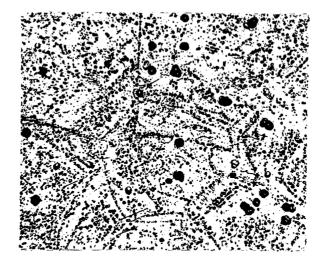


Fig. 21. 2 Atomic Per Cent Pd-U Alloy As Quenched 400° C. Electrolytically etched in 5% Oxalic Acid

2114-7-0





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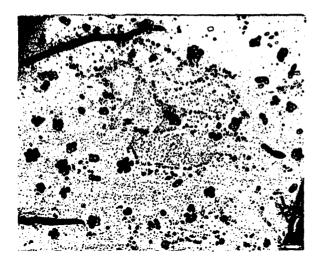
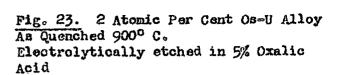


Fig. 22. 2 Atomic Per Cent Os-D Alloy As Cast Electrolytically etched in 5% Oxalic

Acid



2415-1-0

x 250 2415-2-0

x 250

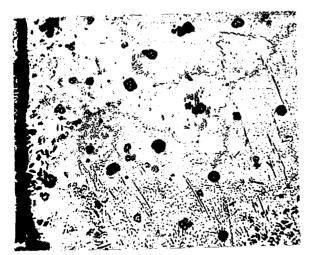


Fig. 24. 2 Atomic Per Cent Os-U Alley As Quenched 800° C. Electrolytically etched in 5% Oxalic Acid

2415-3-0

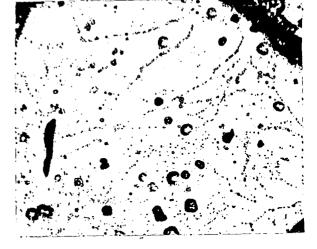


Fig. 25. 2 Atomic Per Cent Os-U Alloy As Quenched 700° C. Electrolytically etched in 5% Omalic Acid

2415-4=0

x 250



~15~



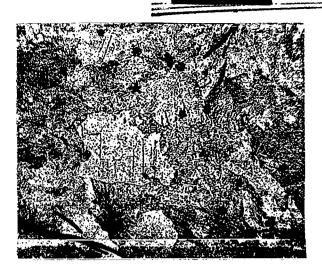


Fig. 26. 2 Atomic Per Cent Os-U Alloy As Quenched 600° C. Electrolytically etched in 5% Oxalic Acid Fig. 27. 2 Atomic Per Cent Os-U Alloy As Quenched 500° C. Electrolytically etched in 5% Oxalic Acid

2415-5-0

x 250

2415=6=0

x 250

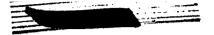


Fig. 28. 2 Atomic Per Cent Os-U Alloy As Quenched 400° C. Electrolytically etched in 2% Oxalic Acid

2415-7-0



-16-



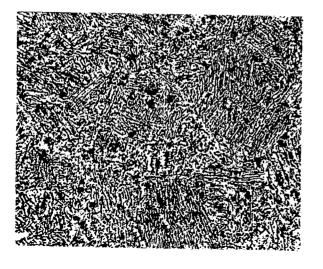




Fig. 29. 2 Atomic Per Cent Ru-U Alloy As Cast Etched in 1;1 Nitric-Acetic Acid

Fig. 30. 2 Atomic Per Cent Ru-U Alloy As Quenched 900° C. Electrolytically etched in 2% Oxalic Acid



Fig. 31. 2 Atomic Per Cent Ru-U Alloy As Quenched 800° C. Electrolytically etched in 5% Oxalic Acid

2416-3-0

2416-2-0

x 250

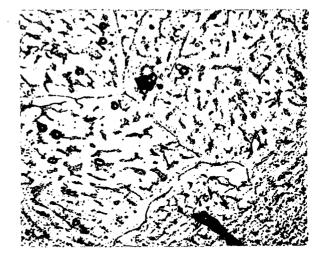


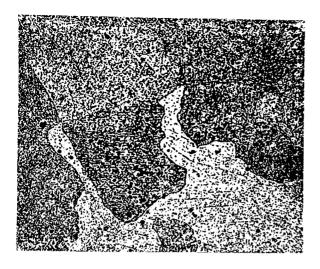
Fig. 32. 2 Atomic Per Cent Ru-U Alloy As Quenched 700° C. Etched in 1:1 Nitric-Acetic Acid

2416-4-0

x 250

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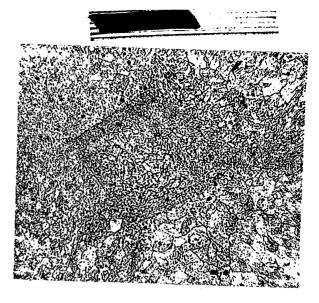


Fig. 33. 2 Atomic Per Cent Ru-U Alloy As Quenched 600° C. Electrolytically etched in 2% Oxalic Acid Fig. 34. 2 Atomic Per Cent Ru-U Alloy As Quenched 500° C. Electrolytically etched in 2% Oxalic Acid

2416-5-0

x 250 2416-6-0

x 250

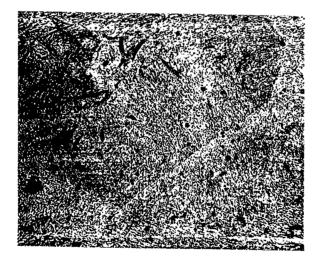
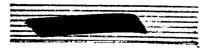


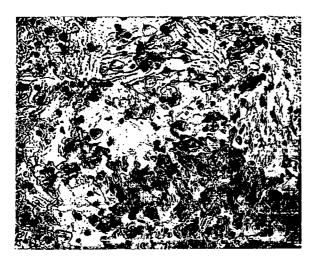
Fig. 35. 2 Atomic Per Cent Ru-U Alloy As Quenched 400° C. Electrolytically etched in 2% Oxalic Acid

2416-7-0

x 250







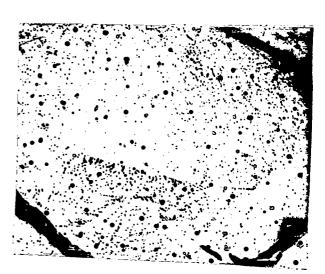


Fig. 37. 2 Atomic Per Cent Ir-U Alloy

Electrolytically etched in 2% Oxalic

Fig. 36. 2 Atomic Per Cent Ir-U Alloy As Cast Electrolytically etched in 10% Chromic

Acid

2420-2-0

Acid

x 250

As Quenched 900° C.

x 250



Fig. 38. 2 Atomic Per Cent Ir-U Alloy As Quenched 800° C. Electrolytically etched in 2% Oxalic Aoid

2420-3-0

2420-1-0

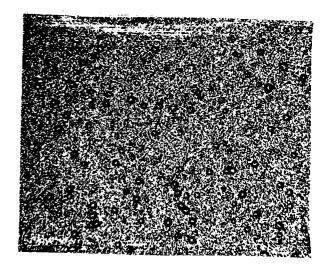


Fig. 39. 2 Atomic Per Cent Ir-U Alloy As Quenched 700° C. Electrolytically etched in 10% Chromic

Electrolytically etched in 10% chronic Acid

2420-4-0

x 250



-19-





Fig. 40. 2 Atomic Per Cent Ir-U Alloy As Quenched 600° C. Electrolytically stched in 2% Oxalic Acid Fig. 41. 2 Atomic Per Cent Ir-U Alloy As Quenched 500° C. Electrolytically etched in 5% Oxalic Acid

2420-5-0

x 250 2420-6-1

x 250

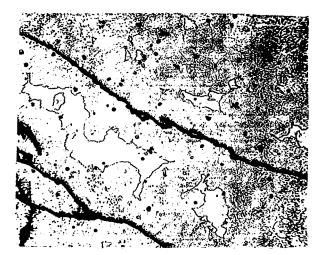
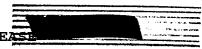


Fig. 42. 2 Atomic Per Cent Ir-U Alloy As Quenched 400° C. Electrolytically etched in 2% Oxalic Acid

2420-7-1

x 250



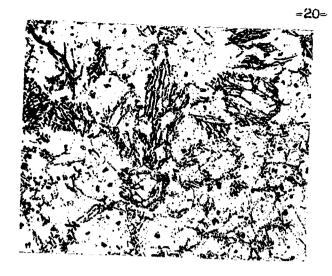




Fig. 43. 2 Atomic Per Cent As Cast Electrolytically etched in 2 Acid	•	Fig. 44. 2 Atomic Per Cent Re As Quenched 900° C. Electrolytically etched in 5% Acid	-
2421-1-0	x 250	2421-2-0	π 250



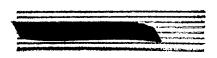
Fig. 45. 2 Atomic Per Cent Re-U Alloy As Quenched 800° C. Electrolytically etched in 5% Oxalic Acid

2421-3-0

Fig. 46. 2 Atomic Per Cent Re-U Alloy As Quenched 700° C. Electrolytically etched in 5% Oxalic Acid

242**1-**4-0

x 250



-21-



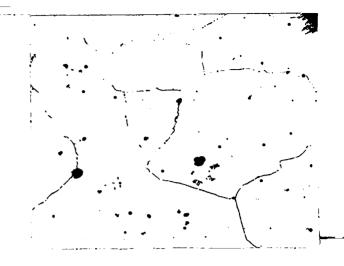
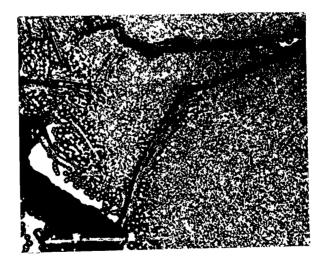


Fig. 47. Atomic Per Cent Re-U Alloy As Quenched 600° C. Electrolytically etched in 2% Oxalic Acid Fig. 48. 2 Atomic Per Cent Re-U Alloy As Quenched 500° C. Electrolytically etched in 2% Oxalic Acid

2421-5-0

2421-6-0

x 250



x 250

Fig. 49. 2 Atomic Per Cent Re=U Alloy As Quenched 400° C. Electrolytically etched in 5% Oxalic Acid

2421-7-0





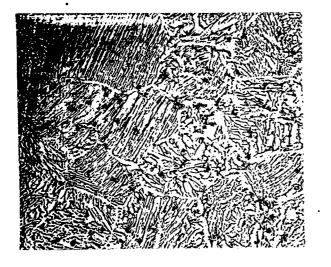


Fig. 50. 2 Atomic Per Cent Rh-U Alloy As Cast Electrolytically etched while polishing.





Fig. 51. 2 Atomic Per Cent Rh-U Alley As Quenched 900° C. Electrolytically etched in 10% Chromic Acid

2422-1-0

x 250 2422-2-0

x 250

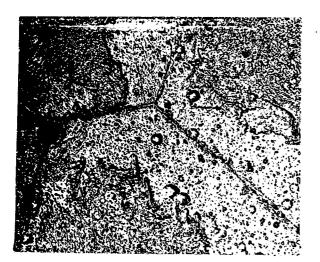


Fig. 52. 2 Atomic Per Cent Rh-U Alloy As Quenched 800° C. Electrolytically stoked in 10% Chromic Acid

2422-3-0

x 250

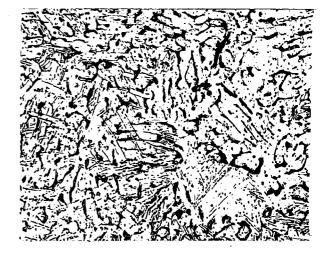
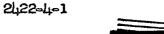


Fig. 53. 2 Atomic Per Cent Rh-U Alloy As Quenched 700° C. Electrolytically etched in 10% Chromic Acid







-23-



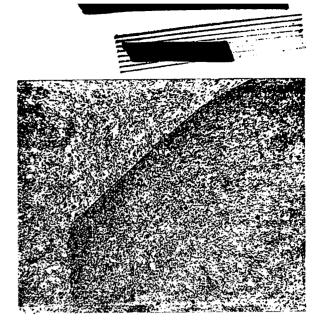


Fig. 54. 2 Atomic Per Cent RheU Alloy As Quenched 600° C. Electrolytically etched in 10% Chromic Acid Fig. 55. 2 Atomic Per Cent Rh-U Alloy As Quenched 500° C. Electrolytically etched in 10% Chromic Acid

2422~5-0 x 250 2422-6-0

x 250

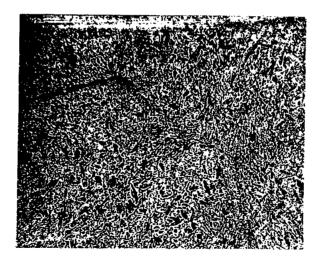


Fig. 56. 2 Atomic Per Cent Rh-U Alloy As Quenched 400° C. Electrolytically etched in 10% Chromic Acid

2422-7-0





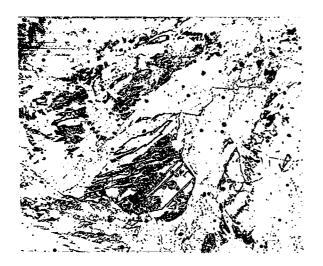
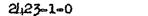




Fig. 57. 2 Atomic Per Cent Ga-U Alloy As Cast Electrolytically stohed in 10% Chromic Acid Fig. 58. 2 Atomic Per Cent Ga-U Alloy As Quenched 900° C. Electrolytically etched in 2% Oxalic Acid



2423-2-0

x 250



Fig. 59. 2 Atomic Per Cent Ga-U Alloy As Quenched 800° C.

Electrolytically etched in 10% Oxalio Acid

2423-3-0

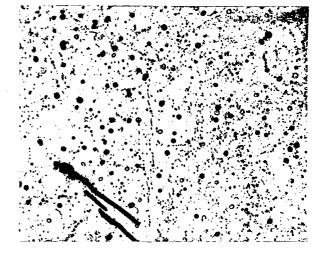


Fig. 60. 2 Atomic Per Cent Ga-U Alloy As Quenched 700° C. Electrolytically etched in 10% Chromic Acid

2423-4-0

x 250

x 250

-23-



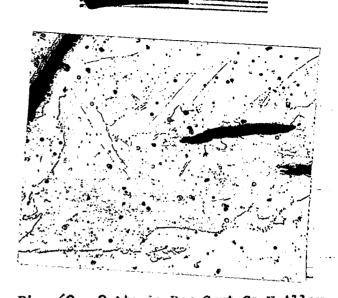


Fig. 61. 2 Atomic Per Cent Ga-U Alloy As Quenched 600° C. Electrolytically etched in 2% Oxalic Acid Fig. 62. 2 Atomic Per Cent Ga-U Alloy As Quenched 500° C. Flectrolytically stched in 10% Chromic Acid

21,23-5-0 x 250 2423-6-0 x 250

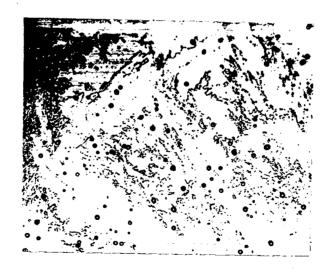


Fig. 63. 2 Atomic Per Cent Ga-U Alloy As Quenched 400° C. Electrolytically etched in 2% Oxalic Acid

2423-7-0

x 2



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